

MASTERARBEIT

Titel der Masterarbeit

"Microbial nitrogen use efficiency along a latitudinal gradient in Western Siberia, Russia"

verfasst von

Anna Knoltsch, BA BSc

angestrebter akademischer Grad

Master of Science (MSc)

Wien, 2013

Studienkennzahl lt. Studienblatt: A 066 833

Studienrichtung lt. Studienblatt: Masterstudium Ökologie UG2002

Betreut von: Univ.-Prof. Dr. Andreas Richter

Content

| I | | Gene | ral Introduction | 5 |
|----|---|-------|--------------------------------------------------------------|-----|
| | 1 | Soil | stoichiometry, community and competition | 5 |
| | 2 | Soil | l inorganic nitrogen | 7 |
| | 3 | Soil | l organic nitrogen | 8 |
| | 4 | Stu | dy aim and working hypothesis | 9 |
| | 5 | Ref | erences | .10 |
| | | | | |
| II | | Manu | script | .17 |
| | 1 | Abs | stract | .17 |
| | 2 | Intro | oduction | .18 |
| | 3 | Mat | terial and methods | .21 |
| | | 3.1 | Sampling sites | .21 |
| | | 3.2 | Sampling design | .23 |
| | | 3.3 | Soil analysis | .24 |
| | | 3.4 | Calculation | .26 |
| | | 3.5 | Statistical analysis | .27 |
| | 4 | Res | sults | .28 |
| | | 4.1 | Carbon and nitrogen pools | .28 |
| | | 4.2 | Soil microbial transformation rates | .30 |
| | | 4.3 | Nitrogen use efficiency | .32 |
| | | 4 4 | Effects of soil parameters on microbial transformation rates | 32 |

| 5 | 5 Dis | cussion | 36 | | | | | | | |
|-----|-------|-----------------------------------------------------|----|--|--|--|--|--|--|--|
| | 5.1 | N transformation rates across latitudinal gradients | 36 | | | | | | | |
| | 5.2 | N transformation rates across depth gradients | 37 | | | | | | | |
| | 5.3 | Nitrogen use efficiency | 38 | | | | | | | |
| 6 | 6 Co | nclusion | 40 | | | | | | | |
| 7 | ' Acl | .cknowledgements | | | | | | | | |
| 8 | B Re | ferences | 42 | | | | | | | |
| III | Zusa | mmenfassung | 48 | | | | | | | |
| IV | Curri | culum vitae | 53 | | | | | | | |

.

I General Introduction

1 Soil stoichiometry, community and competition

Stoichiometric relationships structure all biological systems as primary producers and their consumers depend on certain nutrient ratios to maintain essential functions (Sistla and Schimel, 2012; Kuzyakov and Xu, 2013). Soil is known to control microbial and plant growth due to its' limiting potential in nutrients, mainly phosphorus (P) and nitrogen (N), but also carbon (C) is considered as constraint of microbial cell growth in especially deeper soil layers (Berg and McClaugherty, 2003; Turner et al., 2003; Sistla et al, 2012; Kuzyakov and Xu, 2013). While P is mainly limiting in subtropical and tropical latitudes (Vitousek and Farrington, 1997), arctic and boreal regions are usually poor in N (Hunt et al, 1988; Booth et al, 2005; LeBauer and Treseder, 2008; Sistla et al, 2012). Thus, strong resource competition can be observed among plants and microorganisms (Jones et al, 2013; Kuzyakov and Xu, 2013) but also within the microbial community (Vitousek and Howarth, 1991; Geisseler et al, 2010). Stoichiometric flexibility of biological systems, notably at the organismal scale, is of high advantage (Sistla and Schimel, 2012). Plants may shift their elemental composition (Sterner and Elser, 2002) by, first, changing in physiological state, such as adjustment of growth (Elser et al., 2000) or changes of nutrient ratios in plant tissues e.g. by increasing the carbon to nitrogen ratio (C:N) in woody biomass relative to other tissues (Mack et al, 2004; Sistla and Schimel, 2012) or the foliar nitrogen to phosphorus (N:P) ratios (Elser et al. 2000). Second, biotic re-structuring of plant communities may lead to a shifting dominance among organisms that differ in their average stoichiometry, i.e. shrub expansion in sedge-dominated tussock tundra (Chapin et al, 1986) or grasslands (Sistla and Schimel, 2012).

In general, primary producers show high stoichiometric flexibility because of their capacity to accumulate and store C in different plant tissues. Microbial organisms are much more constrained in their stoichiometry than the soil environment they inhabit (Hessen et al., 2004; Sardans et al., 2011) as prokaryotic microbes lack in nutrient allocation (Sistla and Schimel, 2012). Hence, nutrient alterations in autotroph biomass (Hessen et al., 2004) and soils (Mack et al, 2004) influence easily microbial community composition by shifting the community's C:N (Sistla et al, 2014). N enrichment may select for species with high growth rates and low biomass C:N ratios (Elser et al, 2000). Bacteria are characterized by lower microbial biomass C:N ratio compared to the higher C:N ratio showing fungi (Strickland and Rousk, 2010; Sistla et al., 2014). Hence, increasing N availability in soils may favor bacterial growth, which may lead to greater

bacterial to fungal dominance being accompanied by much faster decomposition rates and rapid nutrient cycling (Weintraub and Schimel, 2005; Sistla and Schimel, 2012). Significant ecosystem C and N loss may select for other species and may occur by e.g. fires - presenting an important component in boreal and grassland ecosystems - because of the relatively low volatilization temperatures of C and N, whilst P is mainly converted into available mineral forms through ash deposition (Eisele et al, 1989). Lower N-availability tends to increase fungal abundances in general, but may also stimulate terrestrial cyanobacteria (Eisele et al, 1989). Fungal dominance may lead to an increase in oxidative enzyme production that causes an elevated decomposition of protected soil organic compounds, being rich in both C and N. Fungal-based food webs are further much more resistant to drought and other climate perturbations than bacterial dominated ecosystems due to their life strategy (Sistla and Schimel, 2012).

Microorganisms developed a wide range of mechanisms to take up inorganic but also small soluble organic molecules (Merrick and Edwards, 1995; Xu et al., 2008; Geisseler et al, 2010) by the use of cell membrane proteins (Geisseler et al, 2010). These proteins may either act as energy consuming transporters that are actively transporting specific molecules from the outside in the cytoplasm of the cell or as passive channels, e.g. linking the transport of ammonia (NH₃) with the symport of H⁺. Ammonium (NH₄⁺) and nitrate (NO₃⁻) were long time considered as main N sources for both bacteria and fungi (Merrick and Edwards, 1995; Geisseler et al, 2010; Kuzyakov and Xu, 2013). Latest research, actually, shows a high preference of microorganisms for amino acids (Kuzyakov and Xu, 2013) and there is evidence that they may also take up small peptides (< 31 amino acids in length) directly (Walker and Altman, 2005; Geisseler et al., 2010; Jones and Kielland, 2012). The transport of peptides and amino acids is performed by transport systems that are bound to the cytoplasmatic membrane (Geisseler et al, 2010). Microbial transport systems for both peptides and amino acids are energy consuming (Anraku 1980) and transport amino acids possessing similar chemical properties.

Plants tend to favour amino acids and peptides as N source (Geisseler et al, 2010) but also nitrate (NO₃⁻) due to its greater mobility in soils and its negative charge (Xu et al., 2008; Jones et al., 2013). Their discrimination for either organic or inorganic components as N source is influenced by temperature and plant species (Xu et al., 2008). Although different plants within a community may have different preferences and needs for organic and inorganic nitrogen forms, similar mechanisms are regulating the N uptake and transport across most species (Näsholm et al., 1998; Persson and Näsholm 2001). However, compared to their microbial antagonists, plants appear to be inferior competitors in inorganic but also organic N uptake (Kuzyakov and Xu, 2013). NO₃⁻ is immobilized twice as fast by microorganisms as by plants, while NH₄⁺ is even

fivefold faster taken up by microorganisms than by the plant community (Jones et al., 2005). Nonetheless, in long term plants do much better due to slower turnover times of roots (Schimel and Bennett, 2004; Jones et al., 2005) and reallocation of N from microorganisms to plants (Kuzyakov and Xu, 2013).

2 Soil inorganic nitrogen

Inorganic nitrogen is only present in small concentrations in soils and turned over very fast. NH₄⁺ is only available in very low concentrations in soils (Jones and Kielland, 2002) and shows especially high depletion in the rhizosphere due to strong root and microbial uptake but very low mobility in the soil solution (Kuzyakov and Xu, 2013). NH₄⁺ is either rapidly taken up and assimilated as microbial (microbial NH₄⁺ immobilization) or plant biomass, or converted to nitrate (NO₃-) during the process of nitrification under aerobic conditions (Booth et al, 2005). In this process, NH₄⁺ may either be directly reduced to nitrite (NO₂⁻) or NO₃⁻ via heterotrophic nitrification (Focht and Verstraete, 1977) or go through the indirect pathway of autotrophic nitrification. During autotrophic nitrification, NH₄⁺ is first converted to hydroxyl amine (NH₂OH) by the enzymatic activity of the ammonium monooxygenase (AMO) (Wood, 1986) and further to NO2 via enzymatic reduction of the hydroxylamine oxidoreductase (HAO) (Yamanaka and Sakano, 1980). A broad range of microorganisms, including archaea and fungi besides bacteria, possess the capability of producing NO₃ through the pathway of nitrification (Focht and Verstraete, 1977). As the produced NO₃ is highly mobile in the soil medium, it serves much more as inorganic N source in soils as NH₄⁺ (Xu et al., 2008). Under anaerobic conditions, NO₃⁻ and NO2 may be further reduced to atmospheric nitrogen (N2) through the intermediates nitric oxide (NO) and/or nitrous oxide (N2O, commonly known as laughing gas) by denitrifying bacteria (Bedmar et al, 2005). In soils, only a small fraction of the soil microbial community is able to denitrify. It is assumed that only five percent of the soil inhabiting bacteria possess denitrifying enzymes, while the denitrifying capabilities of fungi or archaea are still unknown (Henry et al., 2006).

Soil properties such as moisture, pH, temperature and substrate quality are controlling factors of all transformation rates (Nadelhoffer et al, 1992; Chapin, 1996; Cookson et al, 2007). Nitrification rates are especially controlled by ammonium concentrations, while rates of denitrification are additionally controlled by nitrate concentrations (Chapin, 1996). Besides, nitrification rates strongly depend on mineralization rates being proportionally high at decreasing N mineralization

rates. High soil C:N ratios may suppress NO₃ production but promote NO₃ assimilation (Booth et al, 2005).

3 Soil organic nitrogen

The N pool in soils is dominated by organic N forms such as humus, proteins, peptides and amino acids. While amino acids and small peptides may be taken up directly, proteins have to be broken down biologically or abiotically first before microbes are capable of assimilation (Jones and Hodge, 1999; Jan et al., 2009). Most likely the depolymerization of proteins is dominated by extracellular proteolytic enzyme activity instead of the intracellular breakdown by i.e. protozoa (Jan et al., 2009). These exo-enzymes are released by microbial decomposers, mainly mycorrhizal and saprotrophic fungi (Talbot et al., 2008) but also bacteria. With a few exceptions of ericaceous mycorrhizae, saprotrophic fungi generally appear to be the best competitors in the breakdown of proteins (Booth et al, 2005).

Extracellular proteases catalyze the conversion of proteins to peptides and further to free amino acids by hydrolyzing peptide bonds (James et al., 1977) and, thus, accelerate depolymerization processes. As these exo-enzymes are N-rich compounds, the acquisition of extracellular N requires an investment of intracellular N beforehand (Sistla et al, 2012). Thereby, N uptake is inherently linked to nutrient loss for extracellular enzyme production (Mooshammer et al., 2012). Extracellular proteases possess wide substrate-specificities (Gupta et al., 2002; Geisseler et al., 2010) and there is no even distribution of protease expression within the soil microbial community (Fuka et al., 2008; Jan et al., 2009).

Protein depolymerization appears to be significantly slower than the degradation of amino acids themselves (Schimel and Bennett, 2004). Amino acids underlie a very fast turnover with turnover times of 3-6 hours in soils (Kielland et al, 2007) and only 0.3-1.7 hours in leaf litter (Wanek et al., 2010). Thus, amino acids represent a highly dynamic N pool and account for only 5-6% of soil nitrogen, compared to high concentrations of proteinaceous substances that represent around 40% of the soil N (Schulten and Schnitzer, 1997). There are several factors which may influence protein and amino acid turnover besides size and the activity and physiology of the microbial soil community. A broad variety of biotic factors such as temperature, moisture and pH are considered as major controls on these transformation processes (Kang and Lee, 2005). Changes in temperature may lead to limitations in microbial activity at especially very high and low temperatures (Hoyle et al., 2006). Rises in soil moisture cause increases in protein and amino acid degradation processes, but little influence can be observed at high moisture contents

(Tietema et al., 1992). Soil pH and reducing conditions are considered to further highly influence transformation rates. Especially, amino acid turnover correlates negatively with these soil parameters (Marrs et al., 1988). Furthermore, chemical stabilization by soil organic matter (SOM) is of great importance during protein depolymerisation processes. For instance, there are high amounts of humic substances and tannins (polyphenolic substances) in taiga soils, emerging from coniferous wood and foliage. These compounds are able to bind proteins into recalcitrant complexes causing a decreased bioavailability as they are chemically protected against degradation (Schulten and Schnitzer, 1997; Jones and Kielland, 2002; Kraus et al., 2003; Jones and Kielland, 2012). Contrary, it was also observed that humic acids rather stimulated both growth and activity of nitrifying bacteria (Vallini et al., 1997). This was attributed to an increase of microbial membrane permeability due to the surfactant characteristics of humates, allowing a better utilization of nutrients and energy-yielding substrates of the soil einvironment. Thus, humic substances may also act as stimulator for microbial growth and activity in soils (Ganjegunte et al., 2006), although highly depending on the source and amount of present polyphenolic substances (Vallini et al., 1997; Kraus et al., 2003).

4 Study aim and working hypothesis

Understanding the factors that regulate nutrient availability and cycling in soils is essential for generating predictions of consequences of ecosystem alterations including atmospheric carbon dioxide (CO2) and reactive nitrogen (N) enrichments and phosphorus (P) depletions (Kang and Lee, 2005; Craine et al, 2007; Sistla and Schimel, 2012). This study aimed to determine the influence of soil nitrogen availability on microbial nitrogen transformation rates. We expected to find different soil nitrogen availability along major ecosystems as they differ highly in chemical, physical and biological composition. Therefore, we conducted soil analyses along a north-south gradient in Western Siberia, Russia, ranging from arctic tundra to mid-latitude steppe. We specifically focused on protein depolymerization and nitrogen mineralization rates as dominant N transformation rates, on the one hand, and on microbial immobilization rates of amino acids and ammonium on the other one, which were all based on ¹⁵N pool dilution techniques.

5 References

- Anraku, Y (1980): Transport and utilization of amino acids by bacteria. In: Payne JW: Microorganisms and Nitrogen Sources. Wiley, Chichester: 9-33.
- Bedmar EJ, Robles EF, Delgado MJ (2005): The complete denitrification pathway of the symbiotic, nitrogen-fixing bacterium Bradyrhizobium japonicum. Biochemical Society Transactions, 33(1): 141-144.
- Berg B, McClaugherty C (2003): Plant litter: decomposition, humus formation, carbon sequestration. Springer-Verlag, Berlin, Germany.
- Booth MS, Stark JM, Rastetter E (2005): Controls on nitrogen cycling in terrestrial ecosystems: a synthetic analysis of literature data. Ecological monographs, 75(2): 139-157.
- Chapin DM (1996): Nitrogen mineralization, nitrification, and denitrification in a high arctic lowland ecosystem, Devon Island, NWT, Canada. Arctic and alpine research, 28(1): 85-92.
- Chapin III FS, Vitousek PM, Van Cleve K (1986): The nature of nutrient limitation in plant communities. American naturalist, 127 (1): 48-58.
- Craine JM, Morrow C, Fierer N (2007): Microbial nitrogen limitation increases decomposition. Ecology, 88(8): 2105-2113.
- Eisele KA, Schimel DS, Kapustka LA, Parton WJ (1989): Effects of available P and N: P ratios on non-symbiotic dinitrogen fixation in tallgrass prairie soils. Oecologia, 79: 471–474.
- Elser JJ, Sterner RW, Gorokhova E, Fagan WF, Markow TA, Cotner JB, Harrison JF, Hobbie SE, Odell GM, Weider LW (2000): Biological stoichiometry from genes to ecosystems. Ecology Letters, 3: 540–550.
- Focht DD, Verstraete W (1977): Biochemical ecology of nitrification and denitrification. New York, Plenum Press. Advances in Microbial Ecology, 1: 135-214.
- Fuka MM, Engel M, Haesler F, Welzl G, Munch JC, Schloter M (2008): Diversity of proteolytic community encoding for subtilisin in an arable field: spatial and temporal variability. Biology and Fertility of Soily, 45: 181-191.
- Ganjegunte GK, Condron LM, Clinton PW, Davis MR, Mahieu N (2006): Effects of the addition of forest floor extracts on soil carbon dioxide efflux. Biology and fertility of soils, 43(2): 199-207.
- Geisseler D, Horwath WR, Joergensen RG, Ludwig B (2010): Pathways of nitrogen utilization by soil microorganisms a review. Soil Biology and Biochemistry, 42(12): 2058-2067.
- Gupta R, Beg Q, Lorenz P (2002): Bacterial alkaline proteases: molecular approaches and industrial applications. Applied Microbiology and Biotechnology, 59: 15-32.

- Marrs RH, Proctor J, Heaney A, Mountford MD (1988): Changes in soil nitrogen-mineralization and nitrification along an altitudinal transect in tropical rain forest in Costa Rica. The Journal of Ecology 76: 466-482.
- Henry S, Bru D, Stres B, Hallet S, Philippot L (2006): Quantitative detection of the nosZ gene, encoding nitrous oxide reductase, and comparison of the abundances of 16S rRNA, narG, nirK, and nosZ genes in soils. Applied and Environmental Microbiology 72(8): 5181-5189.
- Hessen DO, Ågren GI, Anderson TR, Elser JJ, de Ruiter PC (2004): Carbon sequestration in ecosystems: the role of stoichiometry. Ecology, 85(5): 1179-1192.
- Högberg MN, Briones MJ, Keel SG, Metcalfe DB, Campbell C, Midwood AJ, Thornton B, Hurry V, Linder S, Näsholm T, Högberg P (2010): Quantification of effects of season and nitrogen supply on tree below-ground carbon transfer to ectomycorrhizal fungi and other soil organisms in a boreal pine forest. New Phytologist, 187(2): 485-493.
- Hoyle FC, Murphy DV, Fillery IRP (2006): Temperature and stubble management influence microbial CO2-C evolution and gross N transformation rates. Soil Biology and Biochemistry, 38(1): 71-80.
- Hunt HW, Ingham ER, Coleman DC, Elliott ET, Reid CPP (1988): Nitrogen limitation of production and decomposition in prairie, mountain meadow, and pine forest. Ecology, 69(4): 1009-1016.
- James MN, Hsu IN, Delbaere LT (1977): Mechanism of acid protease catalysis based on the crystal structure of penicillopepsin. Nature, 267(5614): 808-813.
- Jan MT, Roberts P, Tonheim SK, Jones DL (2009): Protein breakdown represents a major bottleneck in nitrogen cycling in grassland soils. Soil Biology & Biochemistry, 41 (11): 2272–2282.
- Jones DL, Hodge A (1999): Biodegradation kinetics and sorption reactions of three differently charged amino acids in soil and their effects on plant organic nitrogen availability. Soil Biology and Biochemistry, 31(9): 1331-1342.
- Jones DL, Kielland K (2002): Soil amino acid turnover dominates the nitrogen flux in permafrost-dominated taiga forest soils. Soil Biology and Biochemistry, 34(2): 209-219.
- Jones DL, Healey JR, Willett VB, Farrar JF, Hodge A (2005): Dissolved organic nitrogen uptake by plants an important N uptake pathway? Soil Biology and Biochemistry, 37(3): 413-423.
- Jones DL, Kielland K (2012): Amino acid, peptide and protein mineralization dynamics in a taiga forest soil. Soil Biology and Biochemistry, 55: 60-69.

- Jones DL, Clode PL, Kilburn MR, Stockdale EA, Murphy DV (2013): Competition between plant and bacterial cells at the microscale regulates the dynamics of nitrogen acquisition in wheat (Triticum aestivum). New Phytologist, 200: 796–807.
- Kang H, Lee D (2005): Inhibition of extracellular enzyme activities in a forest soil by additions of inorganic nitrogen. Communications in Soil Science and Plant Analysis, 36(15-16): 2129-2135.
- Kielland K, McFarland JW, Ruess RW, Olson K (2007): Rapid cycling of organic nitrogen in taiga forest ecosystems. Ecosystems, 10(3): 360-368.
- Kraus TEC, Yu Z, Preston CM, Dahlgren RA, Zasoski RJ (2003): Linking chemical reactivity and protein precipitation to structural characteristics of foliar tannins. Journal of Chemical Ecology, 29: 703-730.
- Kuzyakov Y, Xu X (2013): Competition between roots and microorganisms for nitrogen: mechanisms and ecological relevance. New Phytologist, 198(3): 656-669.
- LeBauer DS, Treseder KK (2008): Nitrogen limitation of net primary productivity in terrestrial ecosystems is globally distributed. Ecology, 89(2): 371-379.
- Leinweber P, Schulten HR (2000): Nonhydrolyzable forms of soil organic nitrogen: Extractability and composition. Journal of Plant Nutrition and Soil Science, 163(4): 433–439.
- Lyyemperumal K, Shi W (2008): Soil enzyme activities in two forage systems following application of different rates of swine lagoon effluent or ammonium nitrate. Applied Soil Ecology, 38: 128-136.
- Mack MC, Schuur EAG, Bret-Harte MS, Shaver GR, Chapin FS III (2004): Ecosystem carbon storage in arctic tundra reduced by long term nutrient fertilization. Nature, 431: 440-443.
- Merrick MJ, Edwards RA (1995): Nitrogen control in bacteria. Microbiological Reviews, 59(4): 604-622.
- Mooshammer M, Wanek W, Schnecker J, Wild B, Leitner S, Hofhansl F, Blöchl B, Hämmerle I, Frank AH, Fuchslueger L, Keiblinger KM, Zechmeister-Boltenstern S, Richter A. (2012): Stoichiometric controls of nitrogen and phosphorus cycling in decomposing beech leaf litter. Ecology, 93(4): 770-782.
- Nadelhoffer KJ, Giblin AE, Shaver GR, Laundre JA (1991): Effects of temperature and substrate quality on element mineralization in six arctic soils. Ecology, 72(1): 242-253.
- Näsholm T, Ekblad A, Nordin A, Giesler R, Högberg M, Högberg P (1998): Boreal forest plants take up organic nitrogen. Nature, 392: 914-916.
- Persson J, Näsholm T (2001): Amino acid uptake: a widespread ability among boreal forest plants. Ecology Letters, 4: 434-438.

- Sardans J, Rivas-Ubach A, Penuelas J (2012): The elemental stoichiometry of aquatic and terrestrial ecosystems and its relationships with organismic lifestyle and ecosystem structure and function: a review and perspectives. Biogeochemistry, 111 (1-3): 1-39.
- Schulten HR, Schnitzer M (1997): The chemistry of soil organic nitrogen: a review. Biology and Fertility of Soils, 26(1): 1-15.
- Sistla SA, Asao S, Schimel JP (2012): Detecting microbial N-limitation in tussock tundra soil: Implications for Arctic soil organic carbon cycling. Soil Biology and Biochemistry, 55: 78–84.
- Sistla SA, Schimel JP (2012): Stoichiometric flexibility as a regulator of carbon and nutrient cycling in terrestrial ecosystems under change. New Phytologist, 196(1): 68-78.
- Sistla SA, Rastetter EB, Schimel JP (2014): Responses of a tundra system to warming using SCAMPS: a stoichiometrically coupled, acclimating microbe-plant-soil model. Ecological Monographs, 84(1): 151-170.
- Sterner RW, Elser JJ (2002): Ecological stoichiometry: the biology of elements from molecules to the biosphere. Princeton, NJ, USA: Princeton University Press.
- Strickland MS, Rousk J (2010): Considering fungal:bacterial dominance in soils-methods, controls, and ecosystem implications. Soil Biology and Biochemistry, 42:1385-1395.
- Talbot JM, Allison SD, Treseder KK (2008): Decomposers in disguise: mycorrhizal fungi as regulators of soil C dynamics in ecosystems under global change. Functional Ecology, 22: 955–963. doi: 10.1111/j.1365-2435.2008.01402.x.
- Tietema A, Warmerdam B, Lenting E, Riemer L (1992): Abiotic factors regulating nitrogen transformations in the organic layer of acid forest soils: moisture and pH. Plant and Soil, 147(1): 69-78.
- Turner BL, Baxter R, Whitton BA (2003): Nitrogen and phosphorus in soil solutions and drainage streams in Upper Teesdale, northern England: implications of organic compounds for biological nutrient limitation. Science of the total environment, 314: 153-170.
- Vallini G, Pera A, Agnolucci M, Valdrighi MM (1997): Humic acids stimulate growth and activity of in vitro tested axenic cultures of soil autotrophic nitrifying bacteria. Biology and fertility of soils, 24(3): 243-248.
- Vitousek PM, Howarth RW (1991): Nitrogen limitation on land and in the sea: how can it occur? Biogeochemistry, 13(2): 87-115.
- Vitousek PM, Farrington H (1997): Nutrient limitation and soil development: experimental test of a biogeochemical theory. Biogeochemistry, 37(1), 63-75.

- Walker JR, Altman E (2005): Biotinylation facilitates the uptake of large peptides by Escherichia coli and other gram-negative bacteria. Applied and environmental microbiology, 71(4): 1850-1855.
- Wanek W, Mooshammer M, Blöchl A, Hanreich A, Richter A (2010): Determination of gross rates of amino acid production and immobilization in decomposing leaf litter by a novel 15N isotope pool dilution technique. Soil Biology & Biochemistry, 42(8): 1293-1302.
- Weintraub MN, Schimel JP (2005): Nitrogen cycling and the spread of shrubs control changes in the carbon balance of arctic tundra ecosystems. BioScience, 55: 408–415
- Wood PM (1986): Nitrification as an bacterial energy source. Prosser, J. I.: 39-62.
- Xu X, Stange CF, Richter A, Wanek W, Kuzyakov Y (2008): Light affects competition for inorganic and organic nitrogen between maize and rhizosphere microorganisms. Plant and Soil, 304(1-2): 59-72.
- Yamanaka T, Sakano Y (1980): Oxidation of hydroxylamine to nitrite catalyzed by hydroxylamine oxidoreductase purified from Nitrosomonas-Europaea. Current Microbiology 4(4): 239-244.

II Manuscript

1 Abstract

Ecosystems show differences in climatic conditions, vegetation and soil organic matter (SOM) content, especially differing in soil N availability along latitudinal gradients. These circumstances require high physiological adaptation of the soil microbial community to compete successfully for nutrients with plants but also with other soil microbial organisms. In this study we aimed at determining the influence of soil N availability on soil microbial transformation rates, focusing on protein depolymerization and N mineralization rates, all based on ¹⁵N pool dilution techniques. Organic and mineral soil samples were taken along a 1,400 km latitudinal transect in Western Siberia, Russia, covering all major ecosystems of tundra, boreal forest, deciduous forest and steppe. N transformation rates seemed to be highly influenced by soil moisture and soil C and N concentrations. Highest protein depolymerization and N mineralization rates occurred in the boreal forest, being accompanied by peaking soil C and N concentrations and high water content, whereas lowest transformation rates were found in tundra and steppe soil. Reduced plant microbial competition for N in deep soil layers was considered to stimulate N mineralization and lower protein depolymerization rates. Highest microbial nitrogen use efficiency (NUE) was found in the southern steppe environment, while values were respectively low at all taiga sites. Unfortunately, NUE could not be calculated for the southern tundra as mineralization rates were under detection limit. High NUE suggested microbial adaptation to high litter C:N ratios, whereas lowest NUE occurred where intermediate litter C:N could be observed. We suggest that initial litter chemistry highly defines microbial NUE, but certainly, there are numerous other factors influencing and changing NUE, e.g. limitations of other nutrients, that should be reconsidered.

Key words: soil microbial transformation rates; N limitation; latitudinal transect; West Siberia

2 Introduction

A new paradigm emerged during the last decade, which characterizes depolymerization of N-containing polymers as main regulator of the overall N cycling (Jones and Kielland, 2002; Schimel and Bennett, 2004; Jan et al, 2009; Fig 1B), while the classical scientific view puts nitrogen mineralization in the center of the terrestrial N cycle (Fig 1A) (Schimel and Benett, 2004). While free amino acids can be taken up directly by the microbial soil community via several membrane transport systems (Jones and Hodge, 1999), polymers are not immediately bioavailable and have to be broken down first to smaller oligomers or monomers (Chapin et al., 2002; Jan et al., 2009). This cleavage is processed by extracellular enzymes (Sistla et al., 2012a) being produced by the soil microbial community. N-containing polymers can be degraded by a range of microbial enzymes, including oxidative enzymes, that are assumed as predominantly N-acquiring enzymes (Talbot et al., 2013; Schnecker et al., 2014) and hydrolytic enzymes, hydrolyzing for instance proteins and peptides (leucine aminopeptidase (LAP)) or chitin (β -1,4-N-acetylglucosaminidase (NAG)) (Sinsabaugh and Shah, 2011). Once the products

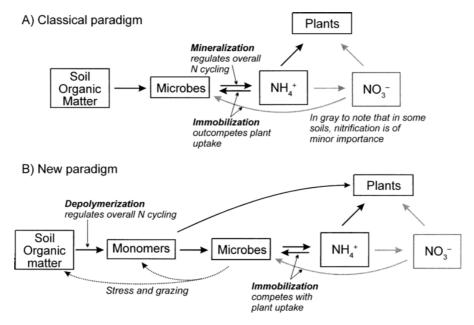


Figure 1: The changing paradigm of the soil N-cycle. (A) The classical paradigm of N-cycling. (B) The new paradigm with depolymerization of N-containing polymers regulating the overall N-cycle. Source: Schimel and Benett, 2004: 594.

of depolymerization enter the dissolved organic nitrogen (DON) pool, they may be taken up and

channeled into microbial metabolism (Fierer et al., 2001), used at higher trophic levels (e.g., micro- and mesofaunal grazing; Elliott et al., 1980) and recycled within the "microbial loop" (Coleman, 1994) or after microbial death (Schimel and Clein, 1996). Depending on the available N sources, which may vary dramatically across soil gradients, the dominating transformation process and main form of N taken up are likely to change (Schimel and Bennett, 2004). N transformation rates and cycling appear to be highly ecosystem specific (Joanisse et al., 2008; Jan et al., 2009), as soils possess great chemical, physical and biological diversity (Jones et al., 2009). Ecosystems along latitudinal gradients show high biological diversity giving rise to variation in litter quality and quantity in a wide range (Xu et al., 2013). Together with different climatic and edaphic conditions, this results in variation in stoichiometry among major ecosystems (Xu et al., 2013), leading to a broad range of soil microbial adaptation (Owen and Jones, 2001; Jones et al., 2009). Xu et al (2013) found strong correlation between latitudinal gradients and soil microbial C:N, suggesting crucial adaptation for microbial N uptake to their soil environment. For example, grassland soils are known to represent N-sufficient conditions (Jan et al., 2009), often even showing inorganic N concentrations comparable to soluble organic ones (Owen and Jones, 2001; Bardgett et al., 2003). Microorganisms living under these circumstances are characterized by high N mineralization rates (Bardgett et al., 2003; Wilkinson et al., 2014) compared to relatively low protein depolymerization rates (Manzoni et al., 2008; Jan et al., 2009), due to N immobilization in the microbial biomass being less important at high bulk N concentrations. Hence, high fractions of N are released back in case of nutrient rich environments through ammonification. Contrary, N-poor soils, as in arctic and boreal ecosystems (Schimel and Bennett, 2004; LeBauer and Treseder, 2008), are considered to be dominated by soluble organic N due to low and slow decomposition and N cycling. As microbial communities are most likely N-limited under such conditions, they retain absorbed N in their biomass (Xu et al., 2013) and only rarely mineralize (Giblin et al., 1991; Schimel and Bennett, 2004). This results in an intense competition for N between plant and microbial community, especially in these low N systems (Jones and Hodge, 1999; Schimel and Bennett, 2004; Xu and Kuzyakov, 2013). Therefore, it is of high advantage to use organic nitrogen forms to decrease dependency on N mineralization and the production of NH₄⁺. Although proteinaceous material such as proteins, peptides and amino acids is dominating the soil nitrogen pool (Schulten and Schnitzer, 1997), it may be protected from microbial attack by physical stabilization or chemical protection reacting with polyphenols e.g. tannins and humic substances (Talbot and Finzi, 2008). Microorganisms, but also plants, may use organic N as N source effectively and compete strongly with soil microbial communities (Nashölm et al., 1998; Schimel and Bennett, 2004; Xu et al., 2006). Thus, soil microsites may highly differ in N availability. Available N, in organic and inorganic form, can diffuse between soil microsites, which may be dominated by either N mineralization or N immobilization (Schimel and Bennett, 2004). Due to microsites that differ strongly in their N availbaility, organisms have to either adapt their physiology or develop mutualistic interactions. Ectomycorrhizal (ECM) fungi may be of importance as they support the depolymerization process of proteins and take up nutrients, especially N, very effectively (Talbot et al., 2013). ECM mycelium in boreal forests even contributes almost 40% to total soil microbial biomass C during summer months (Högberg et al, 2010). Microbial organisms may adapt their physiology by changing their nitrogen use efficiency to sustain their elemental needs (NUE). NUE denotes microbial partitioning of organic N taken up being either incorporated into microbial biomass or released as inorganic N into the environment (Mooshammer et al., 2014a). At high NUE, only small parts of the N taken up, are mineralized and released as ammonium whereas the majority is incorporated into microbial biomass. In contrast, at low NUE only a small fraction of organic N taken up is used for growth while the greater part is mineralized and set free (Mooshammer et al., 2014a).

This project aimed at determining the influence of soil N availability on N transformation rates of the soil microbial community. We hypothesised that higher protein depolymerization rates and amino acid immobilization rates but low nitrogen mineralization rates are found in soils at higher latitudes, having a lower N availability, compared to low-latitude ecosystems. Further, we hypothesized that arctic and boreal soils would exhibit a higher NUE than lower-latitude soils. In contrast to previous studies, which were conducted under laboratory conditions (e.g. Jones and Kielland, 2002; Jan et al., 2009; Jones and Kielland, 2012; Wilkinson et al., 2014), we aimed at characterizing microbial N transformation rates and NUE under field conditions. To achieve this, we sampled soils along a latitudinal gradient from the arctic tundra to the mid-latitude steppe covering all major ecosystems of Western Siberia, Russia.

3 Material and methods

3.1 Sampling sites

We collected soils from seven different sites in Western Siberia, which were located along a 1,400 km latitudinal gradient from the arctic tundra to the mid-latitude steppe (Fig. 1), corresponding to an 8.7°C change in mean annual temperature (Table 1).



Figure 2: Satellite pictures of the latitudinal transect in Western Siberia, © Google earth. Red pins mark the position of the seven different sampling sites, yellow pins the two biggest nearby cities Tazovskiy and Novosibirsk.

The shrubby lichen tundra site (TU) was located approximately 30 km south of the town of Tazovskiy in the southern tundra subzone. The Siberian taiga (boreal forest) was represented by three sites: a *Picea obovata-*forest with high abundances of *Pinus sibirica* and *Vaccinium vitis-idaea* (northern taiga, NT), a boreal coniferous forest with *Sorbus sibirica* and *Pinus sibirica* as most dominant plant species (middle taiga, MT) and a *Picea obovata* forest mixed with *Betula pubescens* and *Abies sibirica* (southern taiga, ST). The forest steppe was characterized by

Table 1: Sampling sites coordinates, elevation, mean annual and August temperature (MAT and Aug_MT), mean annual and August precipitation (MAP and Aug_MP), potential evapotranspiration (PET) and predominant plant species for each site. Climate data: Stolbovoi V and McCallum I (2002), IIASA.

| Site | Latitude | Longitude | Elevation (m) | MAT (°C) | MAP | PET (mm) | Aug_MT | Aug_MP | Dominant plant species* (abundancy in descending order) |
|------|---------------|---------------|---------------|-------------|-------------|-------------|--------|------------|-----------------------------------------------------------------------------------------------------------------------------------------|
| TU | 67°16'20.05"N | 78°50'13.85"E | 30 | -7.6 | (mm) 391 | (mm) 301 | 9 | (mm) 60 | Arctous erythrocarpa, Empetrum nigrum, Betula nana, |
| NT | 63°17'37.54"N | 74°32'9.18"E | 131 | -4.6 | 430 | 405 | 12 | 74 | Cetraria aculeata, Cladonia rangiferina Picea obovata, Pinus sibirica, Vaccinium vitis-idaea, |
| MT | 60° 9'27.08"N | 71°42'57.34"E | 85 | -2.2 | 438 | 490 | 14 | 74 | Pleurozium schreberi, Hylocomium splendens Sorbus sibirica, Pinus sibirica, Abies sibirica, Linnaea |
| ST | 58°17'58.90"N | 68°34'53.71"E | 87 | -0.5 | 396 | 561 | 14 | 68 | borealis, Hylocomium splendens Picea obovata, Betula pubescens, Abies sibirica, Carex |
| FF | 56°14'11.56"N | 70°42'54.90"E | 106 | 0.7 | 340 | 641 | 16 | 59 | macroura, Rubus saxatilis Populus tremula, Inula salicina, Calamagrostis |
| FM | 56°13'54.50 N | 70°43'28.46"E | 102 | 0.7 | 340 | 641 | 16 | 59 | arundinacea, Brachypodium pinnatum, Rubus saxatilis Artemisia macrantha, Calamagrostis epigejos, Vicia cracca, |
| SP | 54°41'41.33"N | 71°38'45.88"E | 72 | 1 | 309 | 700 | 16 | 53 | Thalictrum simplex, Rubus saxatilis Stipa capillata, Festuca valesiaca, Artemisia austriaca, Potentilla bifurca, Artemisia glauca |

^{*} Plant species and abundancy determined by Nikolay Lashchinskiy, Central Siberian Botanical Garden, Siberian Branch of Russian Academy of Sciences, Novosibirsk, Russia

patches of broad leaf hemi-boreal forest (FF) with Populus tremula dominance and dry forest meadow (FM) being dominated by grasses such as Artemisia macrantha and Calamagrostis epigejos. The steppe site (SP) was situated in the mid-latitude Siberian steppe and characterized by the species Stipa capillata, Festuca valesiaca and Artemisia austriaca.

3.2 Sampling design

The sampling was undertaken in August 2012. At each of the seven sites, five (approximately 1m wide) pits were dug to a depth of 100 cm. Samples were taken from the organic soil horizon (O or OA), the mineral topsoil (A or E) and the mineral subsoil horizon (B or E) beneath. In total, 105 soil samples were collected. Soils were classified according to the World Reference Base for Soil Resources (IUSS Working Group WRB, 2006). For basic characterization see Table 2.

Table 2: Physical and chemical parameters of the sampled soils. Values represent means (± standard error).

| Site | Horizon | pH (KCl) | Water content (% FW) | C (mg g ⁻¹ DW) | N (mg g ⁻¹ DW) | C/N (g g ⁻¹) | CaCO ₃ (%) | Soil type according to WRB* |
|------|--------------------|-------------------------------------|----------------------------------------------|-------------------------------------------|----------------------------------------------------------------------|----------------------------------------|----------------------------|----------------------------------------------|
| TU | O A B(C)g | 3.8 (0.1) 3.7 (0.0) 3.9 (0.1) | 63.94 (4.70) 28.37 (0.97) 17.10 (0.50) | 307.9 (37.4) 30.4 (3.0) 4.1 (0.5) | 8.8 (0.7) 1.8 (0.1) 0.4 (0.0) | 34.9 (3.5) 16.4 (0.7) 11.1 (0.8) | - | Turbic Cryosol (thixotropic, reductaquic) |
| NT | Oi AE Bg | AE 3.1 (0.1) 26.27 (1.57) | | 448.4 (7.0) 37.0 (3.1) 8.2 (1.7) | 12.5 (0.3) 1.4 (0.1) 0.5 (0.1) | 35.9 (0.7) 27.4 (2.0) 15.7 (1.5) | - | Histic Podzol (oxyaquic) |
| МТ | Oi A CB | A 3.3 (0.1) 17.33 (1.1) | | 426.1 (24.5) 74.7 (17.3) 16.7 (3.8) | 17.4 (1.0) 3.5 (0.6) 1.0 (0.1) | 24.5 (0.5) 20.8 (1.8) 16.3 (1.7) | - | Endogleyic Regosol |
| ST | Oi A(E) E(A) | (E) 3.6 (0.1) 18.09 (0.70) | | 398.2 (18.3) 43.4 (3.6) 4.8 (0.3) | 15.8 (0.9) 25.4 (0.8) 3.1 (0.2) 14.0 (0.8) 0.5 (0.0) 9.4 (0.2) | | 0.85 (0.00) 1.28 (0.00) | Albic Podzol |
| FF | Oa A B | 6.6 (0.4) 4.3 (0.1) 4.1 (0.0) | 43.76 (2.32) 32.59 (0.89) 32.49 (4.20) | 292.9 (24.1) 45.6 (4.5) 5.2 (0.1) | 17.7 (1.3) 3.6 (0.4) 0.5 (0.0) | 16.5 (0.3) 12.9 (0.2) 10.1 (0.4) | 0.55 (0.14) | Haplic Phaeozeme |
| FM | Oa A Bt | 5.5 (0.3) 4.1 (0.0) 4.0 (0.1) | 35.02 (6.24) 16.03 (8.04) 16.82 (8.00) | 202.1 (22.7) 24.5 (1.6) 5.8 (0.3) | 14.0 (1.6) 1.9 (0.1) 0.5 (0.0) | 14.4 (0.2) 13.0 (0.1) 10.7 (0.2) | 1.05 (0.12) 0.84 (0.00) | Luvic Phaeozeme |
| SP | AO Ak Bk | 4.6 (0.1) 5.1 (0.3) 7.9 (0.4) | 19.01 (0.73) 6.34 (0.38) 6.38 (0.91) | 35.3 (5.4) 20.1 (2.7) 7.2 (0.8) | 3.2 (0.5) 1.8 (0.2) 0.8 (0.1) | 11.0 (0.2) 10.8 (0.3) 9.2 (0.2) | 1.27 (0.17) 7.92 (2.00) | Calcic Kastanozem |

FW, fresh weight; DW, dry weight. * Soil types classified by Norman Gentsch, Institute of Soil Science, Leibniz Universität Hannover, Germany

3.3 Soil analysis

Soil samples were sieved to 2 mm. After removing roots, the soil was used to determine pH, soil water content, water holding capacity (WHC), total carbon (C) and nitrogen concentrations (N), dissolved organic carbon (DOC), total dissolved nitrogen (TDN) and dissolved organic nitrogen (DON), total free amino acids (TFAA), ammonium (NH₄⁺) and nitrate (NO₃⁻) concentrations, gross protein depolymerization and gross amino acid immobilization rates, gross nitrogen mineralization and gross NH₄⁺-immobilization rates.

Soil water content was determined gravimetrically after drying at 60°C. pH-values were measured in 1M KCl-extracts.

3.3.1 Carbon and nitrogen pools

Total organic carbon (C) and nitrogen (N) were measured with an elemental analyzer (EA 1110, CE Instruments) coupled with an isotope-ratio mass spectrometer (Deltaplus, Finnigan MAT, Thermo Fisher) using ground, oven-dried soil samples. Dissolved organic carbon (DOC) and total dissolved nitrogen (TDN) were measured in 1M KCI extracts with a TOC/TN analyzer (TOC-V CPH E200V/TNM-1, Shimadzu). DON was calculated by subtracting NO₃ and NH₄+ from TDN.

*NH*₄⁺ - concentrations were measured colorimetrically by indophenol dye formed by reacting with dichloroisocyanuric acid and salicylate, following a modified protocol after Kandeler and Gerber (1988). NO₃⁻- concentrations were measured photometrically via VCl₃-reduction of nitrate to nitrite and subsequent nitrite detection by dye formation (Miranda KM, Espey MG, Wink DA, 2001). *Total free amino acids (TFAA)* were determined fluorimetrically after reaction with ophthaldialdehyde and 3-mercaptopropionic acid (OPAME) procedure.

3.3.2 Nitrogen fluxes

For determining gross protein depolymerization and amino acid immobilization, gross N mineralization and ammonium immobilization, ¹⁵N-pool dilution assays were performed, as described elsewhere (Wanek et al., 2010; Wild et al., 2013). This technique allows quantification of fluxes by labeling the respective pools with ¹⁵N (i.e., using ¹⁵N amino acids for protein depolymerization and amino acid immobilization, and ¹⁵NH₄+ for N mineralization and ammonium immobilization) and measuring the change in concentration and isotopic enrichment over time (Di et al, 2000).

Gross N mineralization rate (ammonification) was measured by adding 500 μ L of 0.125 mM ($^{15}NH_4$) $_2SO_4$ to soil duplicates (2 g of organic and mineral topsoil, 4 g of mineral subsoil), which

were then incubated for four and 24 hours at 15°C, extracted with 13 mL 2 M KCl, and filtered through ash-free filter paper (Myrold and Tiedje, 1986, modified by Kaiser et al., 2011). NH_4^+ was diffused into acid traps made of Teflon tape enclosing one disc of Whatman filter paper soaked with 10 μ L 2.5 M KHSO₄ (Wanek et al., 2010). Total N and at% ¹⁵N were determined by an elemental analyzer (EA 1110, CE Instruments) coupled with an isotope-ratio mass spectrometer (Deltaplus, Finnigan MAT, Thermo Fisher).

Gross protein depolymerization was measured according to Wanek et al. (2010) with slight modifications to account for the low amino acid concentrations (Wild et al., 2013). 20 µl of ¹⁵Namino acids (mixture of 20 amino acids; concentration of 62.5ng/µl in dest. H₂0), mixed with 0.5 ml (for mineral horizons) to 1 ml (for organic horizons) 10 mM CaSO₄, were added to duplicates of 1 g organic soil or 4 g mineral soil. After incubation for ten or thirty minutes at 15°C, activities were stopped using 20 ml 10 mM CaSO₄ with 3.7% formaldehyde. After centrifuging for five minutes at 10,845 g, samples were filtered through synthetic wool and GF/C filters (Whatman) and transferred to cation exchange cartridges (Dionex OnGuard II H cation exchange cartridges, 057085, Thermo Scientific). Before use, cartridges were rinsed with dest. H₂0, activated with 3 M NH₃ and 1 M HCl and rinsed again with distilled water. Amino acids were eluted with 10 mL 3 M NH₃ and 4 mL dest. H₂0 from cartridges. 10 μl internal standard (mixture of 0.01% norvaline, norleucine, para-chloro-phenylalanine in 0.1 M HCl) were added to the eluate, which was further dried under N₂ (RapidVap N₂ Dry Evaporation System, LABCONCO). Dried samples were taken up in 1.5 ml 20% ethanol and dried in a SpeedVac (SC110 Vacuum Concentrator, Savant). With each batch of samples, amino acid standards and blanks were processed to account for losses due to ion exchange and drying. Dry samples were taken up in 120 µl 0.1 M HCl, 360 µl dest. H₂0 and 320 μl ethanol/pyridine (4:1) and derivatised with 40 μl ethyl chloroformate (ECF) and 800 µl of a 1% ECF in chloroform solution. The organic phase was transferred to GC-vials and dried again in a SpeedVac (SPD131DDA SpeedVac Concentrator, Savant, Thermo Scientific) before being re-dissolved in 50 µl toluol and analyzed with GC-MS (Thermo TriPlus Autosampler, Trace GC Ultra coupled to an ISQ Mass Spectrometer, Thermo Scientific). 2µl of sample were injected in splitless mode at a temperature of 270°C on a PTV-injector, separated on an Agilent DB-5 column with 1 ml/min Helium as carrier gas (GC method: 60°C for 1.5min, ramp 5°C/min to 200°C, ramp 15°C/min to 300°C, 300°C for 4min) and detected in the SIM (Selected Ion Monitoring) mode. Concentrations of alanine, valine, leucine, isoleucine, proline, aspartic acid, glutamic acid and phenylalanine were calculated against external standards. ¹⁵N isotopic compositions were based on peak areas of amino acid fragments as described by Wanek et al., 2010.

3.4 Calculation

Gross protein depolymerization and gross nitrogen mineralization rates were calculated using equation (1), as described by Wanek et al, 2010:

(1)
$$f = \frac{C_2 - C_1}{t} * \frac{\ln \frac{AP_1 - AP_C}{AP_2 - AP_C}}{\ln \frac{C_2}{C_1}}$$

 $f \dots$ Gross flux rate in $\mu g g^{-1}$ dry weight (DW) hours $(h)^{-1}$

t ... Period of incubation in hours (h)

 C_1 , C_2 ... Concentration of either amino acids or NH_4^+ at time 1 and 2 in $\mu g g^{-1} DW$

AP₁, AP₂ ... ¹⁵N-atom percent of either amino acid or NH₄⁺- pools at time 1 and 2

AP_c ... ¹⁵N-atom percent of the unlabeled amino acid or NH₄⁺- pool (natural abundance)

Gross immobilization rates by microorganisms were determined by subsequent equation

(2)
$$i = \frac{C_1 - C_2}{t} * 1 + \frac{\ln \frac{AP_2 - AP_C}{AP_1 - AP_C}}{\ln \frac{C_2}{C}}$$

i ... Gross amino acid immobilization rate in μg g⁻¹ dry weight (DW) hours (h)⁻¹

t ... Period of incubation in hours (h)

C₁, C₂... Concentration either amino acids or ammonium at time 1 and 2 in μg g⁻¹ DW

AP₁, AP₂... ¹⁵N-atom percent of either amino acids or NH₄⁺-Pools at time 1 and 2

 $AP_c \dots ^{15}N$ -atom percent of the unlabeled amino acid or NH_4^+ - pool (natural abundance)

As an indicator for soil microbial nitrogen limitation, *nitrogen use efficiency (NUE)* was calculated based on Wild et al., 2013 and Mooshammer et al., 2014:

$$NUE = \frac{i_{AAs} - f_m}{i_{AAs}}$$

NUE ... Nitrogen use efficiency

 $i_{AAs} \dots$ Gross amino acid immobilization rate in $\mu g \ g^{-1}$ dry weight (DW) hours (h) $^{-1}$

 f_m ... Gross-mineralization rate in μg g^{-1} dry weight (DW) hours $(h)^{-1}$

3.5 Statistical analysis

Statistics were performed in Statgraphics Centurion XVI.I. Data were tested for normal distribution and variance homogeneity. If data did not fit into a normal distributed system, they were \log_{10} or square-root transformed. Significant effects of site or horizon were tested via one-way analysis of variance (One-way ANOVA). Homogenous groups were determined with post hoc Tukey HSD test. Kruskal-Wallis and Mood's Median Tests were performed instead of one-way ANOVA if data didn't show normal distribution and variance homogeneity. Correlation coefficient and its significance were generated via Spearman Rank Correlations since data were not normally distributed. Levels of significance were defined as follows: * p<0.05, ** p<0.01 and **** p<0.001.

4 Results

4.1 Carbon and nitrogen pools

Dissolved organic carbon (DOC) concentrations were significantly different between sites (p < 0.001) and horizon classes (p < 0.001). DOC was the highest in boreal forest sites in organic and mineral soil (Figures 1a-c). In the steppe, DOC increased relatively to N with depth. Tundra and forest steppe sites showed low concentrations of DOC in all soil layers.

Total dissolved nitrogen (TDN) and dissolved organic nitrogen (DON) concentrations showed highly significant differences between soil horizons and sampling site (p < 0.001, respectively). TDN and DON were again highest in boreal sites in all three soil layers (Figures 1a-c). Lowest concentrations were measured in the tundra site, but also in the steppe for the organic and mineral topsoil. Also forest steppe sites showed relatively low concentrations in TDN and DON.

Total free amino acids (TFAA), ammonium (NH₄⁺) and nitrate (NO₃⁻) concentrations differed significantly between horizon classes (TFAA p < 0.001; NH₄⁺ p < 0.01; NO₃⁻ p < 0.05) and sites (TFAA p < 0.001; NH₄⁺ p < 0.05; NO₃⁻ p < 0.001), exceeding about tenfold in organic horizons compared to mineral soil on a dry matter basis (Figures 1d-f). Highest C and N concentrations in organic horizons were again measured in the forest sites. The TFAA pool reached overall highest concentrations in the boreal forest, while tundra and steppe ecosystems were characterized by much lower concentrations. NH₄⁺ reached generally much higher concentrations than NO₃⁻ and concentrations even exceeding TFAA concentrations in the southern taiga sampling site. The steppe was characterized by a very high inorganic N pool (especially in NO₃⁻), which even surpassed the organic TFAA pool.

In mineral topsoil, NO₃ concentrations were again very low at all sites, except of the steppe site showing very high concentrations. TFAA were again the most abundant N form. Highest N pools were found in the forest sites, worth noting the middle and southern taiga sites. Again, the steppe showed much higher concentrations in inorganic N forms instead of organic TFAAs. Mineral subsoil showed a relatively low N pool at all sampling sites. The smallest pool was identified once more in the tundra followed by quite low concentrations at the forest steppe sites. Contrary to the upper soil, highest concentrations occurred in inorganic N forms, notably NH₄⁺. The steppe was again dominated by inorganic NO₃ being followed by NH₄⁺. TFAA-concentrations in the mineral subsoil were in all soils very low and equal to or even beyond inorganic N concentrations.

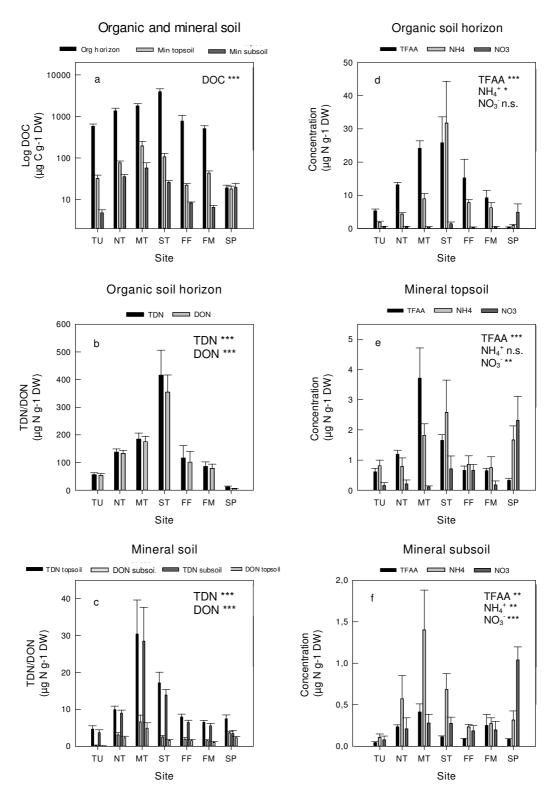


Figure 3: Concentrations of (a) dissolved organic carbon (DOC) for organic and mineral soil, (b) total dissolved nitrogen (TDN) and dissolved organic nitrogen (DON) for organic soil horizon, and (c) for mineral soil; Concentrations of total free amino acids (TFAA), ammonium (NH₄ $^{+}$) and nitrate (NO₃) for (d) organic soil horizon, (e) mineral topsoil and mineral subsoil (f). All bars represent \pm standard error. Levels of significance: ***, p < 0.001; **, p < 0.01; *, p < 0.05; n.s., not significant (one-way ANOVA or Kruskal-Wallis test).

4.2 Soil microbial transformation rates

An ANOVA showed the main effect on gross protein depolymerization to be horizon class not sampling site, whereas the main effect on gross N-mineralization was determined to be the sampling site (see Table 3). Nevertheless, statistical analysis was calculated separated for the three horizon classes organic horizon (O), mineral topsoil (A) and mineral subsoil (M).

Table 3: Analysis of Variance for gross protein depolymerization and gross N mineralization.

| | Gross prote | in depolymerization | N mineralization | | | |
|-------------------|-------------|---------------------|------------------|---------|--|--|
| | F-Ratio | P-Value | F-Ratio | P-Value | | |
| Main effects Site | 2.7 | 0.0196 | 9.2 | 0.0000 | | |
| Horizon class | 10.1 | 0.0000 | 0.2 | 0.9288 | | |

All F-ratios are based on the residual mean square error.

Sampling site differences in gross protein depolymerization, calculated per gram N, could be stated for organic horizon (p < 0.05) and mineral topsoil (p < 0.05), whereas gross N mineralization differed significantly between the sampling sites in all three soil layers (p < 0.01; all transformation rates per g N). Gross protein depolymerization of organic horizon was highest in boreal forest sites (MT, ST, FF), while the lowest rates occurred in the tundra and grassland sites (TU, FS, SP). Mineral topsoil showed no clear pattern in transformation. In mineral subsoil a very high protein depolymerization occurred in the northern taiga (NT), while the other sampling sites showed similar low transformation rates (Figure 2).

Gross N mineralization showed a different pattern in all three soil classes horizons (Figure 2). In the organic horizon, the highest rates were observed in the southern taiga (ST), while the mineralization rates increased from north to south and diminished again from the ST southwards. In mineral topsoil, the N mineralization peaked in the northern taiga (NT) and decreased successively with decreasing latitude, showing higher rates in the boreal sites (CT, ST) than in the hemi-boreal site (FF) and grasslands (FM, SP). In mineral subsoil, a very high transformation occurred in the northern sampling sites compared to the ones in the south. The highest rates were stated in the southern tundra (TU), followed by high rates in the northern (NT) and middle taiga (MT).

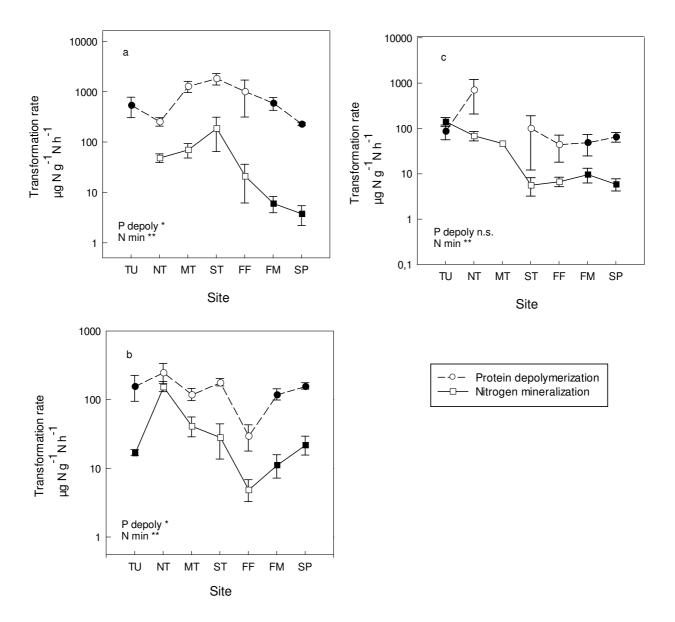


Figure 4: Transformation rates of protein depolymerization (circles) and nitrogen mineralization (squares) at the different sampling sites (black, grassland sites; white, forest sites) for the three soil horizons; organic horizon (a), mineral topsoil (b) and mineral subsoil (c). Note the logarithmic scale. All bars represent \pm standard error. Levels of significance: ***, p < 0.001; **, p < 0.01; *, p < 0.05; n.s., not significant (one-way ANOVA or Kruskal-Wallis test).

4.3 Nitrogen use efficiency

Nitrogen use efficiency (NUE) differed between different sites and horizons (Figure 3). Highest NUE was found in organic horizons and within the organic horizons in the Southern sites. NUE in mineral horizons showed no significant differences, neither in topsoil nor in subsoil. Especially in the subsoil high variation among the replicates of the same sites occurred and made it quite challenging to observe any pattern. Contrary, standard error in the topsoil within each site was quite small, but rates varied highly among the sampling sites. Lowest NUE occurred in the southern taiga, while all other sites showed similar high nitrogen use efficiencies.

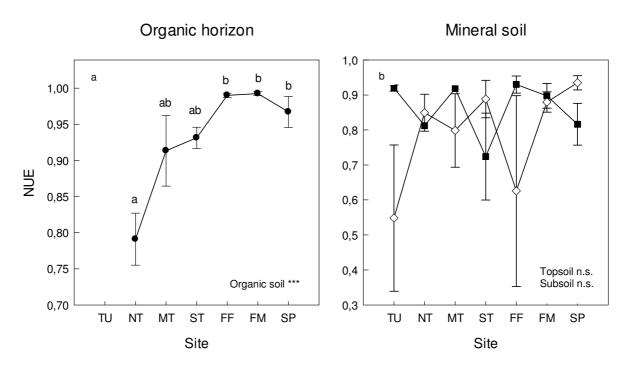


Figure 5: Nitrogen use efficiency (NUE) for (a) organic horizon and (b) mineral topsoil (black squares) and subsoil (white squares) at the seven sampling sites. All bars represent \pm standard error. Levels of significance: ***, p < 0.001; **, p < 0.01; *, p < 0.05; n.s., not significant (one-way ANOVA or Kruskal-Wallis test). Homogenous groups were determined with post hoc Tukey HSD test.

4.4 Effects of soil parameters on microbial transformation rates

Gross protein depolymerization and gross amino acid uptake were correlated significantly to different pools in the organic horizon (Table 2). Generally, all forms of organic (TFAA, DON) and inorganic nitrogen (N, TDN, NH₄⁺) were correlated with both transformation rates. Furthermore, net N mineralization correlated with gross protein depolymerization and amino acid uptake rates. Gross protein depolymerization of mineral soil did not significantly correlate with any soil parameter. Thus, gross amino acid uptake correlated with net protein depolymerization in the

organic horizon, mineral topsoil and mineral subsoil. Gross N mineralization correlated with almost all parameters in the organic horizon and mineral subsoil, whereas in mineral topsoil no correlations could be stated at all. The strongest correlations of gross N mineralization in the organic horizon were found with different soil C and N pools and the soil C/N ratio. Besides, a significantly strong correlation could be found with net N mineralization in this soil horizon. Gross NH₄⁺-uptake correlated similarly with the same parameters but showed also a significant correlation with gross amino acid uptake. In mineral subsoil, gross N mineralization correlated with total soil N, soil C/N ratio, pH-value and WHC. Gross NH₄⁺-uptake correlated negatively with total soil N, WHC, TDN and DON. No other significant correlations with C and N pools could be stated in this horizon.

To investigate soil characteristics and soil microbial transformation rates across sites in the three different soil depths, a Principal Component analysis was performed (Figure 4). In the organic horizon, principal component 1 (PC I) accounted for 45% of variation among soil samples and was positively linked to C and N pools, soil C/N ratio and WHC, while variation in PC II could be explained by differing gross transformation rates.

PC I in mineral topsoil accounted for 36% of variation being positively linked to all C and N pools and WHC, but negatively to NO_3^- and pH. Transformation rates did mainly contribute to variation in component II in mineral topsoil. In mineral subsoil, PC I explained variation for 31% being positively linked to DOC, N pools and pH, but showing a negative link to most gross and net fluxes. PC II accounted for 25% of variation between soil samples and may be explained by depolymerization rates, DOC and NH_4^+ .

Table 2: Spearman Rank Correlation for organic horizon (O), mineral topsoil (A) and mineral subsoil (M) for all sampling sites. The first value represents the correlation coefficient (r); the second value stands for the significance level (p). Black numbers show significant correlations (p < 0.05); grey numbers show not significant correlations (p > 0.05).

| - | Gross pro | otein depoly | merization | Gross a | ımino acid u | ıptake | Net prote | in depolyme | erization | Gross nit | rogen miner | alization | Gross NF | I4+ uptake | | Net nitros | gen minerali | ization |
|----------------------|---------------|--------------|-------------|---------------|---------------|--------------|--------------|---------------|--------------|---------------|----------------|---------------|----------------|---------------|---------------|----------------|---------------|---------------|
| | o | A | M | O | A | M | o^{-1} | A | M | o | \overline{A} | M | O | A | M | 0 | A | M |
| С | 0.2311 | -0.0383 | 0.2027 | 0.2285 | -0.2458 | 0.0677 | -0.1586 | 0.1007 | 0.2271 | 0.7730 *** | -0.0137 | -0.0827 | 0.7592 *** | -0.0013 | -0.3622 | -0.3523 | -0.0170 | -0.1315 |
| N | 0.5608 ** | -0.1415 | 0.1702 | 0.6303 *** | -0.1875 | -0.0369 | -0.2556 | 0.0782 | 0.1499 | 0.4229 * | -0.4160 * | -0.4053 * | 0.4654 * | -0.2810 | -0.4524 * | -0.5538 ** | -0.4797 ** | -0.4485 * |
| C/N | 0.0265 | 0.1444 | 0.0457 | -0.0794 | 0.0549 | 0.0887 | -0.0523 | -0.0012 | 0.1083 | 0.7053 *** | 0.5037 ** | 0.5621 ** | 0.7185 *** | 0.3329 | 0.2868 | -0.2462 | 0.4652 ** | 0.5064 ** |
| pH | 0.0969 | -0.2221 | -0.0057 | 0.1807 | 0.0233 | 0.0358 | -0.2043 | -0.0654 | -0.1219 | -0.5281 ** | -0.5838 *** | -0.3866 * | -0.4725 * | -0.4515 | -0.2620 | 0.0578 | -0.3487 * | -0.3958 * |
| WHC | 0.1920 | -0.0410 | -0.2028 | 0.1640 | -0.0685 | -0.4275 | -0.2236 | 0.0061 | 0.1269 | 0.6417 ** | -0.0313 | -0.4435 * | 0.5746 ** | -0.2026 | -0.5144 ** | -0.1438 | 0.1099 | -0.2332 |
| DOC | 0.5831 ** | 0.2538 | 0.0062 | 0.5987 ** | -0.0659 | 0.0406 | -0.2316 | 0.1691 | 0.3068 | 0.9070 *** | 0.3971 | -0.1262 | 0.9038 *** | 0.4509 | -0.1725 | -0.5862 ** | -0.0882 | -0.2236 |
| TDN | 0.6009 ** | 0.2387 | 0.1316 | 0.6294 *** | -0.2833 | 0.1308 | -0.2205 | 0.2589 | 0.2000 | 0.8831 *** | 0.2978 | -0.3278 | 0.8669 *** | 0.5180 | -0.3913 * | -0.6085 ** | -0.3743 * | -0.4251 * |
| DON | 0.5880 ** | 0.1907 | 0.1480 | 0.6076 ** | -0.2344 | 0.0466 | -0.2191 | 0.1276 | 0.2211 | 0.8981 *** | 0.2403 | -0.3233 | 0.8808 *** | 0.4084 | -0.4863 * | -0.5977 ** | -0.1925 | -0.2890 |
| TFAA | 0.5057 ** | 0.1938 | 0.1575 | 0.5751 ** | -0.1301 | -0.0136 | -0.0763 | 0.1231 | 0.2942 | 0.7115 *** | 0.3424 | -0.0529 | 0.8277 *** | 0.4615 | -0.2759 | -0.6031 ** | -0.1617 | -0.0202 |
| NH4+ | 0.6810 *** | 0.1827 | -0.0819 | 0.7962 *** | -0.0910 | -0.0060 | -0.3468 | 0.3196 | 0.0865 | 0.6130 ** | 0.0405 | -0.2953 | 0.6238 ** | 0.0776 | -0.2756 | -0.5915 ** | -0.4169 | -0.3292 |
| NO3- | -0.0325 | 0.1205 | 0.0820 | 0.0136 | -0.1726 | 0.1685 | -0.0372 | 0.4319 | 0.1798 | -0.0523 | -0.2478 | -0.3391 | 0.0000 | -0.1101 | -0.2795 | -0.2038 | -0.2275 | -0.4538 * |
| Gross p- depoly | | | | 0.8492 *** | 0.5177 ** | 0.5609 * | -0.0719 | 0.3364 | 0.0767 | 0.4002 | 0.1943 | 0.1666 | 0.4116 | 0.1932 | -0.2437 | -0.4704 * | -0.0230 | 0.2602 |
| Gross AA uptake | 0.8492 *** | 0.5177 ** | 0.5609 * | | | | -0.4060 * | -0.5116 ** | -0.5534 * | 0.3982 | -0.0300 | 0.3970 | 0.4421 * | -0.2262 | 0.2561 | -0.5110 * | 0.0623 | 0.1723 |
| Net p-depoly. | -0.0719 | 0.3364 | 0.0767 | -0.4060 * | -0.5116 ** | -0.5534 * | | | | -0.0346 | 0.1362 | -0.1564 | -0.0073 | 0.2200 | -0.2406 | -0.0254 | -0.2138 | 0.0093 |
| Gross N min. | 0.4002 | 0.1943 | 0.1666 | 0.3982 | -0.0300 | 0.3970 | -0.0346 | 0.1362 | -0.1564 | | | | 0.8992 *** | 0.7991 *** | 0.5973 ** | -0.4585 * | 0.1925 | 0.7621 *** |
| Gross NH4+ uptake | 0.4116 | 0.1932 | -0.2437 | 0.4421 * | -0.2262 | 0.2561 | -0.0073 | 0.2200 | -0.2406 | 0.8992 *** | 0.7991 *** | 0.5973 ** | | | | -0.7223 *** | -0.1922 | 0.1552 |
| Net N min. | -0.4704 | -0.0230 | 0.2602 | -0.5110 * | 0.0623 | 0.1723 | -0.0254 | -0.2138 | 0.0093 | -0.4585 * | 0.1925 | 0.7621 *** | -0.7223 *** | -0.1922 | 0.1552 | | | |

Levels of significance: ***, p < 0.001; **, p < 0.01; *, p < 0.05

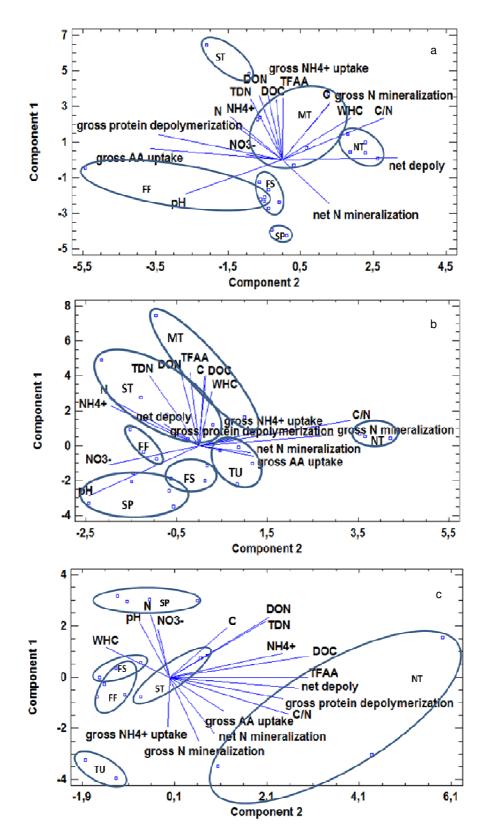


Figure 6: Principal Component Analysis (PCA) for organic horizon (a), mineral topsoil (b) and subsoil (c). Data include all measured pools and fluxes, pH and WHC. TU in organic horizon and MT in mineral subsoil not included due to incomplete data set.

5 Discussion

5.1 N transformation rates across latitudinal gradients

Overall highest protein depolymerisation and N mineralization rates were found in the taiga forest, whereas tundra and steppe were characterized by low transformation rates. These low transformation rates were accompanied by low soil water content. Highest rates occurred in the middle and southern taiga, which were characterized by high soil moisture due to highest annual precipitation along the entire latitudinal transect. These findings suggest that transformation processes highly depend on soil water availability, which may provoke a less active soil microbial community under limited water regimes (Schimel et al., 1996; Booth et al, 2005; Schnecker et al., 2014). Furthermore, high rates in the boreal forest may be due to respectively high vegetation density and high SOM content of soil (Cookson et al., 2007). Several studies already suggested vegetation type (Meyer et al., 2006; Cookson et al., 2007) and SOM content (Ross and Speir, 1979; Booth et al, 2005; Kielland et al, 2007) as important regulators of gross N fluxes and microbial community composition. Other studies argue that major microbial processes are primarily related to C and N availability in soils (Colman and Schimel, 2013). We indeed found highest C and N concentrations in taiga soils and respectively low concentrations in tundra and steppe soils. As a function of arising C and N concentrations, productivity and decomposition processes are stimulated (Christiansen et al., 2012) and may explain the high N fluxes in taiga soils (Ross and Speir, 1979), which may easily exceed those of warmer and more productive ecosystems in the more southern latitudes (Kielland et al, 2007). Boreal ecosystems are further characterized by low pH values which may favour proteolysis in a higher extent than N mineralization (Kielland et al, 2007). This may accelerate protein depolymerization in especially these ecosystems compared to the more southern sampling sites and could explain the observed high fluxes in these soils. The lower protein depolymerization rates in the arctic and the north of the boreal forest may be due to lower temperatures which are considered to reduce the protein turnover (Kielland et al, 2007).

5.2 N transformation rates across depth gradients

Transformation rates were high for both protein depolymerization and N mineralization in the upper organic soil layer. Gross protein depolymerization decreased along the depth gradient which goes in line with former studies stating high turnover rates of amino acids in especially leaf litter (Wanek et al., 2010) and upper soil layers (Kielland et al., 2007; Jones et al., 2009) compared to lower rates in deep soil. This may be due to a decline in microbial biomass with depth (Blume et al, 2002), accompanied by a shift in microbial community composition (Eilers et al., 2012). Hence, microbial soil communities in deep soil may be less active by producing less proteases than microbial communities in upper soil layers (Schnecker et al, 2014), provoking the observed reduction in protein depolymerization rates. Furthermore, deep soils show low abundances in plant biomass being represented only rarely in form of roots. This provides a less competitive environment for microbes in respect of N in deep soil layers and enables a retarded protein depolymerisation. Contrary, in upper soil high microbial and plant biomass leads to respectively high competition and thus nutrient limitation.

While protein depolymerization rates were observed to diminish with increasing soil depth, N mineralization rates increased absolutely but also relatively compared to protein depolymerization rates. As already mentioned, N limitation is assumed to decrease because of reduced competition between microbes and plants and relatively high available amounts of N. Hence, the N demand of microorganisms sinks along depth gradients. Instead of depolymerising the present proteins as N source, they use available compounds just to gain enough C, which is limiting in these soil depths (Nadelhoffer et al., 1991). Hence, N mineralisation occurs where N is more available than needed and not the limiting factor of microbial productivity (Schimel and Bennett, 2004; Booth et al, 2005; Wild et al, 2013). Microbes just take up any source of organic material to compensate for C limitation, mineralizing partly the left over N (Nadelhoffer et al., 1991; Jones and Kielland, 2012). Therefore, amino acids are also taken up and used as energy source. The left over N is mineralized and released as NH₄⁺. This may explain the observed rise of N mineralization rates and the decrease in protein depolymerisation rates with increasing soil depth. High fluctuation in N mineralization with depth occurred especially in the boreal sites, where high rates in the organic horizon diminished rapidly with depth, whereas rates in tundra and meadow soil stayed more constant.

Another reason for the high N mineralization rates in deep soil layers may be the fact that consumption rates may be stimulated by substrate addition (Davidson et al, 1991; Booth et al, 2005). Especially deep soil layers are limited in soil organic matter, and soil microbial activity may be enhanced easily if substrate, here in form of amino acids, is added.

5.3 Nitrogen use efficiency

In contrast to our expectations, nitrogen use efficiency of the soil microbial community decreased with increasing latitude in the organic soil horizon. Rates were of same magnitude in the Southern ecosystems and decreased significantly with intermediate values in the southern and middle taiga and showed lowest values in the northern taiga. In the southern tundra, mineralization rates were that low that they fell below the detection limit. Therefore, NUE could not be calculated for this sampling point, but is expected to be one of the highest along the whole sampling transect as arctic systems are known to represent highly N limited systems (LeBauer and Treseder, 2008; Schimel and Bennett, 2004), wherein microbes have to highly compete for the low available N amounts. Therefore, low N mineralization occurs as only small amounts of N are released back in the nutrient limited environment (Manzoni et al., 2008; Mooshammer et al., 2014). Instead, protein depolymerization is high and the major fraction of available N is incorporated into microbial biomass (Sistla et al, 2012), as could be observed in the organic horizon of the tundra site.

Compared to most studies that state increasing N limitation in soils with decreasing latitudes (e.g. Schimel and Bennett, 2004), our results show a converse pattern. Dry grassland sites were characterized by the highest NUE along this latitudinal gradient showing values of almost 1.0 and indicating a high N limitation within these soils. Plants of temperate grasslands are considered as being even more N constrained than in tundra ecosystems (LeBauer and Treseder, 2008), because of the strong co-limitation between N and water availability (Harpole et al., 2007). Plant productivity in temperate grasslands increases highly in response to both N fertilization and water availability. The soil microbial community may experience the same colimitation in N and water availability because of the generally high temperatures but respectively low precipitation in the grassland ecosystems (see Table 1, p. 23). These circumstances may cause reduced microbial activity to save energy costs and result in high microbial N immobilization but low N mineralization to use all available N to form new biomass. Thus, microbial organisms living within these ecosystems exert a need for high NUE (Sterner and Elser, 2002). Furthermore, average senesced leaf C:N ratios within grassland ecosystems are respectively high with C:N ratios of 60.9 compared to the average C:N ratio of 52.9 across all biomes (Yuan and Chen, 2009). Highest leaf C:N in this study was stated for tundra ecosystems with a C:N ratio of 72.8, which may provoke high microbial N immobilization and low mineralization rates as we could observe in the southern tundra. This proposes high microbial NUE within these low C:N tundra ecosystems.

In contrast, sites showing more N-sufficient conditions, generated by low substrate C:N ratios, establish soil microbial communities which mineralize excess N and lower their N immobilization rates. This results in a low microbial NUE of at least 0.68 in organic soil horizons (Mooshammer et al., 2014) in C:N low ecosystems such as tropical forests, showing senesced leaf C:N ratios of 41.9. Boreal forests are also characterized by relatively low senesced leaf C:N ratios of 48.2 (Yuan and Chen, 2009). Organisms living within these conditions are expected to adapt to their environment (Mooshammer et al., 2014), resulting in an intermediate NUE which was observed in our study. Microbial soil community in the boreal forest ecosystem, especially the northern taiga, showed the lowest NUE along the latitudinal gradient, with an NUE of 0.8 ±0.05. Thus, N immobilization and mineralization rates in organic soil horizons seem to depend mostly on initial litter chemistry and not on climatic regimes (Manzoni et al., 2008).

Mineral topsoil did not show any significant differences in NUE across the latitudinal gradient. Our results are supported by other studies which proposed equal N limitation across different ecosystems (Elser et al, 2007; Jones et al., 2009). This may be due to the fact that organic horizons are mostly influenced by the given environment, climate regime and vegetation, while mineral subsoil is much more isolated and thereby enables more constant conditions. Physical and chemical parameters seem to be more important and major controls on nutrient cycling in deeper soil layers. Especially N mineralization in deep soil horizons suggests high dependency on different N pools, pH and WHC.

6 Conclusion

N transformation rates change with latitude and seem to be influenced not only by climate variables such as temperature and soil moisture, but also by soil C and N concentration, SOM content and pH. Highest protein depolymerization and N mineralization rates occurred in the boreal forest, being accompanied by peaking soil C and N concentrations. Tundra and grassland sites were characterized by much lower transformation rates. Transformation rates were high for both protein depolymerization and N mineralization in the upper organic soil layer. Reduced plant microbial competition for N in deep soil layers may stimulate N mineralization and lower protein depolymerization rates. Contrary to our expectations, soil microbial NUE at temperate grassland sites was characterized by highest values reaching almost 1.0, while values were respectively low at all taiga sites. Unfortunately, NUE could not be calculated for the southern tundra as mineralization rates were under detection limit. High NUE suggested microbial adaptation to high litter C:N ratios, whereas lowest NUE occurred where intermediate litter C:N was observed. Thus, NUE seems to depend mostly on initial litter chemistry. Certainly, there are numerous other factors influencing and changing NUE e.g. the limitation of other nutrients besides N itself which may lead to a decrease in NUE (Mooshammer et al., 2014). Furthermore, enzymes production involves mainly N investment and may have a strong impact on NUE (Schimel and Weintraub, 2003). So definitely, there are much more relevant factors that differentially effect soil community physiology and the microbial N cycle.

7 Acknowledgements

First of all, I thank my family for supporting me all over the last twenty five years. Without my parents and grandmothers' encouragement it wouldn't have been possible to achieve the entire education that I was lucky to experience.

Furthermore, I want to thank all my colleagues from the Division Terrestrial Ecosystem Research, who helped me in any way during my Master thesis. Here, I especially thank Mag. Birgit Wild, who incessantly had time to help me with any matter of concern and who was the best scientific support I can think of. I also like to thank Mag. Florian Hofhansl for having always an open ear and a solution for any technical and statistical concern. Great thanks to Univ.-Prof. Dr. Andreas Richter who enabled this master thesis and all affiliated experiences and acquaintances.

Last but not least, I thank the members of the CRYOCARB - team for their cooperation during field work, especially Mag. Norman Gentsch, who classified the soil types after the World Reference Base.

8 References

- Bardgett RD, Streeter TC, Bol R (2003): Soil microbes compete effectively with plants for organic-nitrogen inputs to temperate grasslands. Ecology, 84(5): 1277-1287.
- Blume E, Bischoff M, Reichert J, Moorman T, Konopka A, Turco R (2002): Surface and subsurface microbial biomass, community structure and metabolic activity as a function of soil depth and season. Applied Soil Ecology, 20: 171-181.
- Booth MS, Stark JM, Rastetter E (2005): Controls on nitrogen cycling in terrestrial ecosystems: a synthetic analysis of literature data. Ecological monographs, 75(2): 139-157.
- Chapin III FS, Matson PPA, Mooney H (2002): Principles of terrestrial ecosystem ecology. Springer, New York, New York, USA.
- Christiansen CT, Svendsen SH, Schmidt NM, Michelsen A (2012): High arctic heath soil respiration and biogeochemical dynamics during summer and autumn freeze-in effects of long-term enhanced water and nutrient supply. Global Change Biology, 18: 3224–3236.
- Coleman DC (1994): The microbial loop concept as used in terrestrial soil ecology studies. Microbial Ecology, 28 (2): 245-250.
- Colman BP, Schimel JP (2013): Drivers of microbial respiration and net N mineralization at the continental scale. Soil Biology and Biochemistry, 60: 65-76.
- Cookson WR, Osman M, Marschner P, Abaye DA, Clark I, Murphy DV, Stockdale EA, Watson CA (2007): Controls on soil nitrogen cycling and microbial community composition across land use and incubation temperature. Soil Biology and Biochemistry, 39(3): 744-756.
- Davidson EA, Hart SC, Shanks CA, Firestone MK (1991): Measuring gross nitrogen mineralization, and nitrification by 15 N isotopic pool dilution in intact soil cores. Journal of Soil Science, 42(3): 335-349.
- Di HJ, Cameron KC, McLaren RG (2000): Isotopic dilution methods to determine the gross transformation rates of nitrogen, phosphorus, and sulfur in soil: a review of the theory, methodologies, and limitations. Australian Journal of Soil Research, 38: 213-230.
- Eilers KG, Debenport S, Anderson S, Fierer N (2012): Digging deeper to find unique microbial communities: The strong effect of depth on the structure of bacterial and archaeal communities in soil. Soil Biology and Biochemistry, 50: 58-65.
- Elliott ET, Anderson RV, Coleman DC, Cole, CV (1980): Habitable pore space and microbial trophic interactions. Oikos, 35 (3): 327-335.
- Elser JJ, Bracken ME, Cleland EE, Gruner DS, Harpole WS, Hillebrand H, Ngai JT, Seabloom EW, Shurin JB, Smith JE (2007): Global analysis of nitrogen and phosphorus limitation of

- primary producers in freshwater, marine and terrestrial ecosystems. Ecology letters, 10(12): 1135-1142.
- Fierer N, Allen AS, Schimel JP, Holden PA (2003): Controls on microbial CO2 production: a comparison of surface and subsurface soil horizons. Global Change Biology 9: 1322-1332.
- Giblin AE, Nadelhoffer KJ, Shaver GR, Laundre JA, McKerrow AJ (1991): Biogeochemical diversity along a riverside toposequence in arctic Alaska. Ecological Monographs, 61(4): 415-435.
- Harpole WS, Potts DL, Suding KN (2007): Ecosystem responses to water and nitrogen amendment in a California grassland. Global Change Biology, 13(11): 2341-2348.
- Högberg MN, Briones MJ, Keel SG, Metcalfe DB, Campbell C, Midwood AJ, Thornton B, Hurry V, Linder S, Näsholm T, Högberg P (2010): Quantification of effects of season and nitrogen supply on tree below-ground carbon transfer to ectomycorrhizal fungi and other soil organisms in a boreal pine forest. New Phytologist, 187(2): 485-493.
- IUSS Working Group WRB (2006): World Soil Resources Reports No. 103. FAO, Rome, Italy; http://www.fao.org/fileadmin/templates/nr/images/resources/pdf_documents/wrb2007_red .pdf, 24.10.2013.
- Jan MT, Roberts P, Tonheim SK, Jones DL (2009): Protein breakdown represents a major bottleneck in nitrogen cycling in grassland soils. Soil Biology & Biochemistry, 41 (11): 2272–2282.
- Joanisse GD, Bradley RL, Preston CM, Bending GD (2009): Sequestration of soil nitrogen as tannin–protein complexes may improve the competitive ability of sheep laurel (Kalmia angustifolia) relative to black spruce (Picea mariana). New Phytologist, 181(1): 187-198.
- Jones DL, Hodge A (1999): Biodegradation kinetics and sorption reactions of three differently charged amino acids in soil and their effects on plant organic nitrogen availability. Soil Biology and Biochemistry, 31(9): 1331-1342.
- Jones DL, Kielland K (2002): Soil amino acid turnover dominates the nitrogen flux in permafrost-dominated taiga forest soils. Soil Biology and Biochemistry, 34(2): 209-219.
- Jones DL, Kielland K (2012): Amino acid, peptide and protein mineralization dynamics in a taiga forest soil. Soil Biology and Biochemistry, 55: 60-69.
- Jones DL, Kielland K, Sinclair FL, Dahlgren RA, Newsham KK, Farrar JF, Murphy DV (2009): Soil organic nitrogen mineralization across a global latitudinal gradient. Global Biogeochemical Cycles, 23(1), DOI: 10.1029/2008GB003250.
- Jones DL, Owen AG, Farrar J (2002): Simple method to enable the high resolution determination of total free amino acids in soil solutions and soil extracts. Soil Biology & Biochemistry, Journal 34: 1893-1902.

- Kaiser C, Fuchslueger L et al. (2011): Plants control the seasonal dynamics of microbial N cycling in a beech forest soil by belowground C allocation. Ecology, 92(5): 1036-1051.
- Kandeler E, Gerber H (1988): Short-term assay of soil urease activity using colorimetric determination of ammonium. Biology and fertility of Soils, 6: 68-72.
- Kerley SJ Read DJ (1995): The biology of mycorrhiza in the Ericaceae. New Phytologist, 131: 369–375. doi: 10.1111/j.1469-8137.1995.tb03073.x.
- Kielland K, McFarland JW, Ruess RW, Olson K (2007): Rapid cycling of organic nitrogen in taiga forest ecosystems. Ecosystems, 10(3): 360-368.
- LeBauer DS, Treseder KK (2008): Nitrogen limitation of net primary productivity in terrestrial ecosystems is globally distributed. Ecology, 89: 371-379.
- Manzoni S, Jackson RB, Trofymow JA, Porporato A (2008): The global stoichiometry of litter nitrogen mineralization. Science, 321: 684-686.
- McGroddy ME, Daufresne T, Hedin LO (2004): Scaling of C: N: P stoichiometry in forests worldwide: Implications of terrestrial redfield-type ratios. Ecology, 85, 2390-2401.
- Meyer H, Kaiser C, Biasi C, Hämmerle R, Rusalimova O, Lashchinsky N, Baranyi C, Daims H, Barsukov P, Richter A (2006): Soil carbon and nitrogen dynamics along a latitudinal transect in Western Siberia, Russia. Biogeochemistry, 81(2): 239-252.
- Miranda KM, Espey MG, Wink DA (2001): A rapid, simple spectrophotometric method for simultaneous detection of nitrate and nitrite. Nitric Oxide, Feb 5(1):62-71.
- Mooshammer M, Wanek W, Hämmerle I, Fuchslueger L, Hofhansl F, Knoltsch A, Schnecker J, Takriti M, Watzka M, Wild B, Keiblinger KM, Zechmeister-Boltenstern S, Richter A (2014a): Adjustment of microbial N use efficiency to C:N imbalances regulates soil N cycling. Nature Communications, 5. doi:10.1038/ncomms4694.
- Mooshammer M, Wanek W, Zechmeister-Boltenstern S, Richter A (2014b): Stoichiometric imbalances between terrestrial decomposer communities and their resources: mechanisms and implications of microbial adaptations to their resources. Frontiers in microbiology, 5.
- Myrold DD, Tiedje JM (1986): Simultaneous estimation of several nitrogen-cycle rates using N-15- theory and application. Soil Biology & Biochemistry, 18(6): 559-568.
- Nadelhoffer KJ, Giblin AE, Shaver GR, Laundre JA (1991): Effects of temperature and substrate quality on element mineralization in six arctic soils. Ecology, 72(1): 242-253.
- Nashölm T, Ekblad A, Nordin A, Giesler R, Högberg M, Högberg P (1998): Boreal forest plants take up organic nitrogen. Nature, 392: 914-916.
- Owen AG, Jones DL (2001): Competition for amino acids between wheat roots and rhizosphere microorganisms and the role of amino acids in plant N acquisition. Soil Biology and Biochemistry, 33(4): 651-657.

- Schimel JP, Bennett J (2004): Nitrogen mineralization: challenges of a changing paradigm. Ecology, 85(3): 591-602.
- Schimel JP, Clein JS (1996): Microbial response to freeze-thaw cycles in tundra and taiga soils. Soil Biology and Biochemistry, 28 (8), 1061-1066.
- Schimel JP, Weintraub MN (2003): The implications of exoenzyme activity on microbial carbon and nitrogen limitation in soil: a theoretical model. Soil Biology and Biochemistry, 35: 549-563.
- Schnecker J, Wild B, Hofhansl F, Alves RJE, Bárta J, Čapek P, Fuchslueger L, Gentsch N, Gittel A, Guggenberger G, Hofer A, Kienzl S, Knoltsch A, Lashchinskiy N, Mikutta R, Šantrůčková H, Shibistova O, Takriti M, Urich T, Weltin G and Richter A (2014): Microbial community composition and enzyme activities in cryoturbated arctic soils are controlled by environmental parameters rather than by soil organic matter properties. PLoS One, 9(4): e94076. doi:10.1371/journal.pone.0094076.
- Schulten HR, Schnitzer M (1997): The chemistry of soil organic nitrogen: a review. Biology and Fertility of Soils, 26(1): 1-15.
- Sinsabaugh RL, Shah JJF (2011): Ecoenzymatic stoichiometry of recalcitrant organic matter decomposition: the growth rate hypothesis in reverse. Biogeochemistry, 102(1-3): 31-43.
- Sistla SA, Asao S, Schimel JP (2012): Detecting microbial N-limitation in tussock tundra soil: Implications for Arctic soil organic carbon cycling. Soil Biology and Biochemistry, 55: 78–84.
- Sistla SA, Schimel JP (2012): Stoichiometric flexibility as a regulator of carbon and nutrient cycling in terrestrial ecosystems under change. New Phytologist, 196: 68.78.
- Sterner RW, Elser JJ (2002): Ecological Stoichiometry: The Biology of Elements from Molecules to the Biosphere. Princeton University Press.
- Stolbovoi V, McCallum I (2002): Land resources of Russia (CD). International Institute for Applied Systems Analysis and the Russian Academy of Science, Laxenburg, Austria; http://webarchive.iiasa.ac.at/Research/FOR/russia cd/index.htm, 21.9.2013.
- Talbot JM, Bruns TD, Smith DP, Branco S, Glassman SI, Erlandson S, Vilgalys R, Peay KG (2013): Independent roles of ectomycorrhizal and saprotrophic communities in soil organic matter decomposition. Soil Biology and Biochemistry, 57: 282-291. doi:10.1016/j.soilbio.2012.10.004.
- Talbot JM, Finzi AC (2008): Differential effects of sugar maple, red oak, and hemlock tannins on carbon and nitrogen cycling in temperate forest soils. Oecologia, 155(3): 583-592.
- Vitousek PM, Gosz JR, Grier CC, Melillo JM, Reiners WA, Todd RL (1979): Nitrate losses from disturbed ecosystems. Science, 204 (4392), 469-474.

- Wanek W, Mooshammer M, Blöchl A, Hanreich A, Richter A (2010): Determination of gross rates of amino acid production and immobilization in decomposing leaf litter by a novel 15N isotope pool dilution technique. Soil Biology & Biochemistry, 42(8): 1293-1302.
- Wild B, Schnecker J, Bárta J, Čapek P, Guggenberger G, Hofhansl F, Kaiser C, Lashchinsky N, Mikutta R, Mooshammer M, Šantrůčková H, Shibistova O, Urich T, Zimovj SA, Richter A (2013): Nitrogen dynamics in Turbic Cryosols from Siberia and Greenland. Soil Biology and Biochemistry, 67: 85-93. doi:10.1016/j.soilbio.2013.08.004.
- Wild B, Schnecker J, Knoltsch A, Takriti M, Mooshammer M, Gentsch N, Mikutta R, Eloy Alves RJ, Gittel A, Lashchinsky N, Richter A (in review): Microbial nitrogen limitation along a latitudinal transect in Western Siberia: No decrease from high to low latitudes, but from organic to mineral soil horizons.
- Wilkinson A, Hill PW, Farrar JF, Jones DL, Bardgett RD (2014): Rapid microbial uptake and mineralization of amino acids and peptides along a grassland productivity gradient. Soil Biology and Biochemistry, 72: 75-83.
- Xu X, Ouyang H, Kuzyakov Y, Richter A, Wanek W (2006): Significance of organic nitrogen acquisition for dominant plant species in an alpine meadow on the Tibet plateau, China. Plant and Soil, 285(1-2): 221-231.
- Xu X, Thornton PE, Post WM (2013): A global analysis of soil microbial biomass carbon, nitrogen and phosphorus in terrestrial ecosystems. Global Ecology and Biogeography, 22: 737–749.
- Yuan ZYY, Chen HYH (2009): Global trends in senesced-leaf nitrogen and phosphorus. Global Ecology and Biogeography 18: 532-542.

III Zusammenfassung

Ökosysteme weißen nicht nur Unterschiede in ihren klimatischen Bedingungen und ihrer Vegetation auf sondern auch in ihrer Bodenzusammensetzung in Form von organischen und mineralischen Bodensubstanzen. Hohe Variation ist vor allem in der Verfügbarkeit diverser Nährstoffe, wie beispielsweise Kohlenstoff (C) und Stickstoff (N), zu beobachten. Vor allem variiert die Nährstoff-Verfügbarkeit in Böden stark entlang latitudinaler Gradiente. Dies setzt eine hohe physiologische Anpassungsfähigkeit der bodenbewohnenden Mikroorganismen voraus, um eine hohe Konkurrenzstärke gegenüber Pflanzen aber auch anderen Bodenorganismen zu allem Prozessraten gewährleisten. Hierbei spielen vor eine grundlegende Rolle. Mikroorganismen können Aminosäuren direkt aufnehmen, um ihren Stickstoffbedarf zu decken. Liegen Aminosäuren allerdings als längere Peptide, Proteine oder anderwärtige Polymere vor, so geht eine obligate Protein-Depolymerisierung der mikrobiellen Aufnahme voraus. Ist die Stickstoff-Verfügbarkeit im Boden gering, werden die depolymerisierten Aminosäuren von der mikrobielle Bodengemeinschaft augenblicklich aufgenommen und als Biomasse assimiliert (mikrobielle Aminosäuren-Immobilisierung). Aufgenommene Aminosäuren können auch wieder ausgeschieden und in Form von Ammonium (NH₄⁺) in den Boden freigesetzt werden (Stickstoff-Mineralisierung). Dieser freigesetzte Stickstoff kann folglich wieder aufgenommen, assimiliert NH_4^+ Immobilisierung) und wieder freigesetzt (mikrobielle werden. Transformationsraten variieren in ihrer Höhe je nach dominanter Stickstoff Quelle sowie genereller Stickstoff Verfügbarkeit des jeweiligen Bodens. Die daraus resultierende mikrobielle Anpassungsfähigkeit kann mittels Stickstoff Nutzungseffizienz (NUE) angegeben werden. Die NUE drückt aus wieviel des aufgenommen Stickstoffs in die mikrobielle Biomasse eingebaut beziehungsweise als NH₄⁺ wieder freigesetzt wird. Je höher sie ist, umso mehr wird der N von der mikrobiellen Gemeinschaft genützt und als Biomasse assimiliert, d.h. die mikrobielle Aminosäuren-Immobilisierung überwiegt gegenüber der Stickstoff-Mineralisierung.

Ziel dieser Arbeit war die Analyse des Einflusses der Stickstoff-Verfügbarkeit auf mikrobielle Prozessraten in Böden. Hierbei standen Raten der Protein-Depolymerisierung, der Stickstoff-Mineralisierung sowie der mikrobiellen Immobilisierung von Aminosäuren und Ammonium (NH₄+) im Fokus des Forschungsinteresses. Alle Raten wurden mittels "¹⁵N pool dilution" analysiert. Diese Methode erlaubt die Quantifizierung von Flüssen indem betreffende Pools mit dem schwereren Stickstoff Isotop ¹⁵N markiert werden und Konzentrationsveränderungen sowie Veränderungen in der Isotopen Anreicherung über die Zeit gemessen werden. So werden

beispielsweise ¹⁵N-Aminosäuren zur Bestimmung der Protein-Depolymerisierung und Immobilisierung von Aminosäuren, beziehungsweise ¹⁵NH₄⁺ zur Bestimmung der Stickstoff-Mineralisierung und Immobilisierung von NH₄⁺ eingesetzt. Hierfür wurden organische und mineralische Bodenproben entlang eines 1400km langen latitudinalen Transekts in West Sibirien, Russland, gezogen. Dadurch wurde sichergestellt, dass alle wesentlichen Ökosysteme abgebildet wurden: die von Permafrost beherrschte Tundra, der boreale Nadelwald der Taiga, der laubwerfende Mischwald und die Steppe.

Zusammenfassend lässt sich feststellen, dass N-Transformationsraten von Klimavariablen wie Temperatur und Bodenfeuchtigkeit, aber auch von C und N Konzentrationen in Böden sowie der höchsten Bodensubstanzen sind. Die Menge organischer abhängig Protein Depolymerisierungsraten und N Mineralisierungsraten wurden im borealen Nadelwald gemessen, begleitet von den höchsten C und N Konzentrationen in diesen Böden. Tundra und Steppe waren hingegen durch niedrigere Transformationsraten gekennzeichnet. Transformationsraten waren generell in allen organischen Horizonten hoch. Eine Abnahme in der Protein Depolymerisierung aber eine Zunahme in der N Mineralisierung mit steigender Bodentiefe lässt auf eine reduzierte Konkurrenz zwischen Pflanzen und Mikroorganismen in tieferen Bodenschichten schließen.

Im Gegensatz zu unserer Annahme, wurde die höchste Stickstoff Nutzungs Effizienz (NUE) in den südlichen Ökosystemen der Steppe und Waldsteppe gemessen, während eine signifikante Abnahme in der mikrobiellen NUE mit zunehmendem geographischen Breitengrad beobachtet werden konnte. Die NUE in der Tundra konnte aufgrund der geringen Mineralisierungsraten, welche unter das Detektionslimit fielen, nicht kalkuliert werden. Die vorgefundene hohe NUE im Süden weißt auf eine starke mikrobielle Adaptierung an das hohe C:N Verhältnis von Streu- und Laubfall hin, während Ökosysteme mit mittleren C:N Verhältnissen durch niedrige NUE Werte gekennzeichnet waren. Die ursprüngliche chemische Zusammensetzung des Streueintrags scheint demnach der bestimmende Faktor der mikrobiellen NUE zu sein. Dennoch sind unzählige weitere Faktoren, wie beispielsweise die Limitierung anderer Nährstoffe, als wesentliche Faktoren in der Entwicklung und Veränderung der mikrobiellen NUE zu berücksichtigen.

IV Curriculum vitae

Personal Information

| Name | Knoltsch, Anna |
|----------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Nationality | Austria |
| Education | |
| 1995 – 1999 | Elementary school Theodor-Körnerschule, Klagenfurt |
| 1999 – 2007 | Secondary school Mössingerstraße, Klagenfurt |
| Jan – June 2006 | Pupil exchange at the Johann Goethe Schule, St.Petersburg, Russian Federation |
| June 2007 | Higher education entrance qualification |
| 2007 – 2010 | Bachelor of Political Sciences, University of Vienna |
| 2007 – 2011 | Bachelor of Biology with specification in Ecology, University of Vienna |
| Feb 2011 - July 2011 | ERASMUS-exchange at the Université Montpellier II, Montpellier, France |
| 2010-2013 | Master of Ecology, specification in "Ecosystem Sciences", University of Vienna Master thesis: "Microbial nitrogen use efficiency along a transect in Western Siberia", Department of Microbiology and Ecosystem Science, Division of Terrestrial Ecosystem Research |

Working experience

April 2010 – June 2010 Voluntary service in soil field mapping

BFW - Federal Office of Wood, Vienna

August 2010 Internship in herbarising

Botanical Garden, Klagenfurt

September 2010 Internship in limnology

KIS - Carinthian Institute for lake research, Klagenfurt

April - June 2011 Determination of the macrozoobenthos in the lagoon of

Thau

Université Montpellier II, Montpellier, France

2011 – 2013 Tutor of the practical course "Pflanzenanatomie"

Department of Molecular System Biology, University of

Vienna

Summer term 2012 - 2014 Tutor of the field course "Kenntnis mitteleuropäischer

Lebensgemeinschaften"

Faculty of Life Sciences, University of Vienna

Winter term 2012 Tutor of the course "Chemische Methoden der Ökologie"

Department for microbiology and ecosystem research,

Division of Terrestrial Ecosystem Research, University of

Vienna

Scientific publications, posters and talks

Mooshammer et al. (2014): Adjustment of microbial N use efficiency to C:N imbalances regulates soil N cycling. Nature Communications, 5. doi:10.1038/ncomms4694.

Schnecker et al. (submitted September 2014): Enzyme patterns in topsoil and subsoil horizons along a latitudinal transect in Western Siberia.

- Schnecker et al. (2014): Effects of Soil Organic Matter Properties and Microbial Community Composition on Enzyme Activities in Cryoturbated Arctic Soils. PLoS One 9(4): e94076. doi:10.1371/journal.pone.0094076.
- Schnecker et al. (2014): Microbial community composition and enzyme activities in cryoturbated arctic soils are controlled by environmental parameters rather than by soil organic matter properties. EGU April 2014, Vienna. EGU2014-5430.
- Schnecker et al. (2014): Enzyme activities along a latitudinal transect in Western Siberia. EGU April 2014, Vienna. EGU2014-10595. Poster.
- Takriti et al. (2013): Carbon and nitrogen interactions along a North-South transect in Western Siberia. CRYOCARB Workshop March 2013, České Budějovice. Poster.
- Takriti et al. (2014): Substrate use efficiency of microbial communities along a latitudinal transect through Western Siberia. EGU April 2014, Vienna. EGU2014-15031.
- Wild et al. (2014): Cycling of organic and mineral nitrogen along a latitudinal transect in Western Siberia. EGU April 2014, Vienna. EGU2014-5870.
- Wild et al. (in review): Microbial nitrogen limitation along a latitudinal transect in Western Siberia: No decrease from high to low latitudes, but from organic to mineral soil horizons. Global Biogeochemical Cycles. 2014GB004914.